

## Lesson Quiz

# Observing Chemical Change

Write the letter of the correct answer on the line at the left.

1. \_\_\_\_ Which of the following is true about chemical reactions?
  - A They are accompanied by changes in energy.
  - B They form new substances with new properties.
  - C both A and B
  - D neither A nor B
2. \_\_\_\_ In an endothermic reaction, energy is
  - A absorbed
  - B released
  - C converted to mass
  - D synthesized
3. \_\_\_\_ Which of the following is NOT a physical property?
  - A melting point
  - B state of matter
  - C density
  - D flammability
4. \_\_\_\_ Substances formed as a result of a chemical reaction are called
  - A catalysts
  - B precipitates
  - C products
  - D reactants

If the statement is true, write *true*. If the statement is false, change the underlined word or words to make the statement true.

5. \_\_\_\_\_ In an exothermic reaction, products have more energy than reactants.
6. \_\_\_\_\_ Water boils at 100°C. This is an example of a chemical property.
7. \_\_\_\_\_ Substances that enter into a chemical reaction are called products.
8. \_\_\_\_\_ The ability to react with oxygen is an example of a chemical property.
9. \_\_\_\_\_ Another name for a chemical change is a chemical bond.
10. \_\_\_\_\_ In a physical change, some of the physical properties of the substance may be altered and the chemical composition remains the same.

# Chapter 1study Guide Energy And Chemical Change Answers

**Arshad Iqbal**



## **Chapter 1 study Guide Energy And Chemical Change Answers:**

*Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual* Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter.

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