Exercises

 An equation in the model shows the fusion of two deuterium nuclei to form a nucleus of tritium. Suggest another product that might form in this reaction.

2. Describe how to find the identity of the species X in the equation

To solentify the element, Subtract (so from 92
235
 =0 $U+1$ an \rightarrow 152 =0 $Nd+X+3$ for.
92-60=32 Next, Subtract the 152+1 from 235 to subtract the Ge isotrope Germanium has an atomic number of 32. $235-(152+1)=82$ $\times=\frac{82}{52}$ Ge

3. What is missing in the following reaction?

4. An atom of U-235 absorbs a neutron and produces an atom of Sb-125 and four neutrons. Identify the other nuclide formed in this reaction. Write the equation to support your answer.

$$92-51=41$$
 niobium-106 $235 U + {}^{1}_{0} N \rightarrow {}^{125}_{51} Sb + {}^{10}_{41} Nb + 4 {}^{1}_{0} N$

5. Identify the following equations as fission or fusion.

	Fission or Fusion?
$^{2}_{1}H + ^{2}_{1}H \rightarrow ^{3}_{1}H + ^{1}_{1}p$	Fusion
$^{235}_{92}\text{U} + ^{1}_{0}n \rightarrow ^{141}_{56}\text{Ba} + ^{92}_{36}\text{Kr} + 3 ^{1}_{0}n$	Fission
$^{138}_{92}U + ^{1}_{0}n \rightarrow ^{138}_{54}Xe + ^{95}_{38}Sr + 3 ^{1}_{0}n$	Fission
$^{3}_{2}\text{He} + ^{3}_{2}\text{He} \rightarrow ^{4}_{2}\text{He} + 2^{1}_{1}\text{H}$	Fusion

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