

Chapter 2 The Nature of Matter — Answers

2.1 Properties of Matter

Warm Up, p. 60

1. For example: cold, white, crystalline, slippery, compressible, cohesive
2. For example: light, heavy, fluffy, wet
3. For example: the air temperature and pressure

Quick Check, p. 62

1. The stuff that materials are composed of
2. A quality of a thing, especially a quality common to a group, type, class, etc.
3. A quality that is or depends upon the amount of the material
4. A property that describes a chemical change. That is one in which a new substance(s) or species is formed, or otherwise describes the tendency of a chemical to react

Quick Check, p. 63

1. The average mechanical energy of the particles that compose a material
2. The total mechanical energy of an object's or a material's particles
3. The energy transferred from one body to another because of a difference in temperature

Quick Check, p. 65

1. The process of changing from a solid to a liquid
2. The vigorous bubbling that occurs within the body of a liquid as it vaporizes internally
3. The amount of heat required to melt a specified amount of a substance at its melting point

Quick Check, p. 67

1. A property of a solution that only depends on the concentration of the dissolved particles, not on their identity
2. For example: boiling point elevation
3. A solution of a metal and another solid or solids

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