

## Chapter 2

### I. Atoms and Molecules

- A. Emergent properties – atoms and molecules
- B. Life requires about 25 chemical elements
- C. Elements combine to form compounds - 2 or more elements in fixed ratio
  - 1. table salt is NaCl; iodized (fig. 2.4)
  - 2. Water
- D. Atoms consist of protons, neutrons and electrons
  - 1. atom is the smallest unit of matter
  - 2. subatomic particles make up the atom.
  - 3. Differences in elements
  - 4. isotopes - different numbers of neutrons so mass changes
- E. Electron arrangement determines chemical properties of atom
  - 1. electrons determine how an atom behaves
  - 2. electrons vary in energy
  - 3. electrons occur at certain energy levels
- F. Ionic bonds are attractions between ions of opposite charges
  - 1. ion = loss or gain of electron
  - 2. ions with opposite charges attract each other
  - 3. resulting compound is electrically neutral
  - 4. sodium and chloride ions are always present in 1:1 ratio
- G. Covalent bonds = sharing of electrons in outer shell
  - 1. molecule
  - 2. single covalent bond
  - 4. double bond
  - 5. H<sub>2</sub> and O<sub>2</sub> are molecules but not compounds
- H. Chemical reactions rearrange matter
  - 1. chemical reactions
  - 2. reactants
  - 3. product
  - 4.  $2\text{H}_2 + \text{O}_2 = \text{H}_2\text{O}$

### II. Properties of water

- A. water is a polar molecule
- B. Water's polarity leads to H bonding
- C. Hydrogen bonds make liquid water cohesive
  - 1. H bonds last for only a few trillionths of a second
  - 2. cohesion and adhesion
  - 3. surface tension
- D. Hydrogen bonds of water moderate temperature
  - 1. heat
  - 2. temperature
- E. Ice is less dense than water
- F. water is a versatile solvent
  - 1. solution
  - 2. solvent
  - 3. solute
  - 4. aqueous solution

### III. pH Scale

- A. The chemistry of life is sensitive to acid and basic conditions
  - 1. some water molecules break apart naturally in water
  - 2. ions formed are called hydrogen (H<sup>+</sup>) and hydroxide (OH<sup>-</sup>) ions
  - 3. for life chemistry, proper balance of these ions is necessary
  - 4. the more acidic a solution, the more H<sup>+</sup> ions. Acids are substances that donate H<sup>+</sup>
  - 5. in your stomach, HCl donates H<sup>+</sup> ions when in solution

# Chapter 2 Basic Chemistry Worksheet Answers

**S Marginson**



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**Chapter Resource 2 Chemistry of Life Biology** Holt Rinehart & Winston,Holt, Rinehart and Winston Staff,2004

**Concepts of Earth Science & Chemistry Parent Lesson Plan** John Hudson Tiner,2013-08-26 Concepts of Earth and Chemistry Course Description This is the suggested course sequence that allows one core area of science to be studied per semester You can change the sequence of the semesters per the needs or interests of your student materials for each semester are independent of one another to allow flexibility Semester 1 Earth Blending a creationism perspective of history with definitions of terms and identification of famous explorers scientists etc this book gives students an excellent initial knowledge of people and places encouraging them to continue their studies in depth Semester 2 Chemistry Chemistry is an amazing branch of science that affects us every day yet few people realize it or even give it much thought Without chemistry there would be nothing made of plastic there would be no rubber tires no tin cans no televisions no microwave ovens or something as simple as wax paper This book presents an exciting and intriguing tour through the realm of chemistry as each chapter unfolds with facts and stories about the discoveries of discoverers Find out why pure gold is not used for jewelry or coins Join Humphry Davy as he made many chemical discoveries and learn how they shortened his life See how people in the 1870s could jump over the top of the Washington Monument Exploring the World of Chemistry brings science to life and is a wonderful learning tool with many illustrations and biographical information [Educart CBSE Class 12 Chemistry One Shot Question Bank 2026 \(Includes PYQs for 2025-26\)](#) , **The Online Tutor's Toolkit** Molly Bolding,2022-10-12 This book contains everything you need to know to get started as an online tutor It covers the essentials of tutoring choosing your tech and software managing homework and getting set up alongside detailed guidance focusing on each level of tuition With techniques developed through research and first hand experience the author explains exactly how to turn existing subject knowledge into effective tutoring for students of all ages in a variety of subjects Divided into two parts the first answers the logistical questions facing every new tutor such as what equipment do I need Where can I apply How much should I charge The second half focuses on how to tutor different age groups effectively and subject specific areas including English Maths and Science as well as the author s tried and tested 5 step process for choosing a subject assessing a student and planning their first lessons There is also information on how to support students writing personal statements and applying to university as well as teaching English as a Second Language Alongside tailored up to date information on available software hardware exam specifications and the online tutoring marketplace the book contains a 10 week timetable of adaptable lesson plans so new tutors can get started immediately Finally there are two additional downloadable chapters which expand on less common subjects and another which includes a digital download of every resource from the book With suggestions for resources homework and timings to support you at every stage this is an essential read for anyone wanting to succeed as an online tutor *Merrill Chemistry* Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 **The General Education**

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**Prentice Hall Science Explorer** Michael J. Padilla, 2002  
Publishing, Bowker, 1999-12      **The Arithmetic Teacher**, 1987      **Children's Books in Print** R R Bowker  
Guidelines for Mine Waste Dump and Stockpile Design  
Mark Hawley, 2017-04 Guidelines for Mine Waste Dump and Stockpile Design is a comprehensive practical guide to the investigation design operation and monitoring of mine waste dumps dragline spoils and major stockpiles associated with large open pit mines These facilities are some of the largest man made structures on Earth and while most have performed very well there are cases where instabilities have occurred with severe consequences including loss of life and extensive environmental and economic damage Developed and written by industry experts with extensive knowledge and experience this book is an initiative of the Large Open Pit LOP Project It comprises 16 chapters that follow the life cycle of a mine waste dump dragline spoil or stockpile from site selection to closure and reclamation It describes the investigation and design process introduces a comprehensive stability rating and hazard classification system provides guidance on acceptability criteria and sets out the key elements of stability and runout analysis Chapters on site and material characterisation surface water and groundwater characterisation and management risk assessment operations and monitoring management of ARD emerging technologies and closure are included A chapter is also dedicated to the analysis and design of dragline spoils Guidelines for Mine Waste Dump and Stockpile Design summarises the current state of practice and provides insight and guidance to mine operators geotechnical engineers mining engineers hydrogeologists geologists and other individuals that are responsible at the mine site level for ensuring the stability and performance of these structures Readership includes mining engineers geotechnical engineers civil engineers engineering geologists hydrogeologists environmental scientists and other professionals involved in the site selection investigation design permitting construction operation monitoring closure and reclamation of mine waste dumps and stockpiles      *Books in Print Supplement*, 2002      **Index to Media and Materials for the Mentally Retarded, Specific Learning Disabled, Emotionally Disturbed** National Information Center for Special Education Materials, 1978      **Exploring Earth and Space** Michael DiSpezio, 1995 A textbook exploring such aspects of matter and energy as heat electricity and nuclear chemistry with suggested activities and review questions at the end of each chapter      **Organic Chemistry Laboratory** Charles E. Bell, 1997      *The Software Encyclopedia 2001*, 2001

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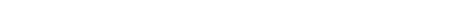
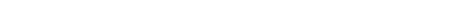
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