

## Concentration and Molarity PhET Labs

Name: \_\_\_\_\_

**Procedure:** [https://phet.colorado.edu/sims/html/concentration/latest/concentration\\_en.html](https://phet.colorado.edu/sims/html/concentration/latest/concentration_en.html)

### Part 1: Dissolution and Saturation

Take some time to play and familiarize yourself with the simulation. Click on everything. Move all the sliders. Notice what happens to the concentration as solid solute is added and when evaporation occurs.



Concentration

- How does the concentration change as solid solute is added? concentration increases
  - How does the concentration change as additional water is added? concentration decreases
  - How does the concentration change as evaporation occurs? concentration increases
  - How do you know when a solution is saturated? concentration does not change (it also shows on screen)
  - When a solution is saturated, and additional solid solute is added, what happens? solid does not dissolve and settles at bottom
  - Why do you think this is? solvent can't hold any more solute
  - How does adding this additional solute change the concentration of this saturated solution? decreases concentration
  - How does evaporation change the concentration of a saturated solution? no change- solution is already saturated
- Part 2: Concentrated Solutions (same website as part 1 just click on the Solution option)**
- Adding a concentrated solution ... describe a way to determine the concentration of the solution in the spigot. Write your plan here: empty the container and add only one solution from the dropper and then use the sensor to measure concentration



- Using your plan ... how might you get that concentrated solution to become saturated? you can evaporate
- Does your plan work for all the other solutions too? yes Why? / Why Not? evaporation reduces the amount of water (or solute) so concentration increases



### Part 3: Molarity [https://phet.colorado.edu/sims/html/molarity/latest/molarity\\_en.html](https://phet.colorado.edu/sims/html/molarity/latest/molarity_en.html)

**Molarity (M) =**  $\frac{\text{moles of solute (mol)}}{\text{volume of solution (L)}}$

Molarity is moles per liter, that is, how many moles of solute (entire salt) is dissolved per liter of solution.

First, determine the saturation concentration of each of the solutions, that is, how concentrated can you get each solution before the solution is saturated. If you can't determine the concentration using the simulation "Molarity", try using the simulation "Concentration" (You will use this information again in Part 5, if your instructor requires it)



Molarity

Cobalt (II) nitrate	<u>5.640 M</u>	Potassium chromate	<u>3.350 M</u>
Cobalt chloride	<u>4.330 M</u>	Nickel (II) chloride	<u>5.210 M</u>
Potassium dichromate	<u>0.510 M</u>	Copper sulfate	<u>1.380 M</u>
Gold (III) chloride	<u>2.30 M</u>	Potassium permanganate	<u>0.480 M</u>

# Concentration And Molarity Post Lab Exercises Answers

**L Cohen**



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**Laboratory Manual for Biotechnology and Laboratory Science** Lisa A. Seidman, Mary Ellen Kraus, Diana Lietzke Brandner, Jeanette Mowery, 2022-12-23 Provides the basic laboratory skills and knowledge to pursue a career in biotechnology Written by four biotechnology instructors with over 20 years of teaching experience it incorporates instruction exercises and laboratory activities that the authors have been using and perfecting for years These exercises and activities help students understand the fundamentals of working in a biotechnology laboratory Building skills through an organized and systematic presentation of materials procedures and tasks the manual explores overarching themes that relate to all biotechnology workplaces including forensic clinical quality control environmental and other testing laboratories Features Provides clear instructions and step by step exercises to make learning the material easier for students There are Lab Notes for Instructors in the Support Material see tab below Emphasizes fundamental laboratory skills that prepare students for the industry Builds students skills through an organized and systematic presentation of materials procedures and tasks Updates reflect recent innovations and regulatory requirements to ensure students stay up to date Supplies skills suitable for careers in forensic clinical quality control environmental and other testing laboratories **Laboratory Manual for Biotechnology** Verma, Ashish S./ Das Surajit & Singh Anchal, Laboratory Manual in Biotechnology Students **Methods in Biotechnology** Seung-Beom Hong, M. Bazlur Rashid, Lory Z. Santiago-Vázquez, 2016-08-01 As rapid advances in biotechnology occur there is a need for a pedagogical tool to aid current students and laboratory professionals in biotechnological methods Methods in Biotechnology is an invaluable resource for those students and professionals Methods in Biotechnology engages the reader by implementing an active learning approach provided advanced study questions as well as pre and post lab questions for each lab protocol These self directed study sections encourage the reader to not just perform experiments but to engage with the material on a higher level utilizing critical thinking and troubleshooting skills This text is broken into three sections based on level Methods in Biotechnology Advanced Methods in Biotechnology I and Advanced Methods in Biotechnology II Each section contains 14 22 lab exercises with instructor notes in appendices as well as an answer guide as a part of the book companion site This text will be an excellent resource for both students and laboratory professionals in the biotechnology field Foundation Course for NEET (Part 2): Chemistry Class 9 Lakhmir Singh & Manjit Kaur, Our NEET Foundation series is sharply focused for the NEET aspirants Most of the students make a career choice in the middle school and therefore choose their stream informally in secondary and formally in senior secondary schooling accordingly If you have decided to make a career in the medical profession you need not look any further Adopt this series for Class 9 and 10 today Essential Methods of Instrumental Analysis Frank M. Dunnivant, Jake W. Ginsbach, 2024-08-05 Intuitively organized textbook aligned to common analytical instrumentation courses for undergraduate students Through an analytical approach Essential Methods of Instrumental Analysis provides an expansive

overview of common instruments and methods and their applications for undergraduate students integrating experimental protocols with real result examples to deliver a well rounded understanding of the inner workings of the instruments and enabling students to evaluate the success of their experiments and create scientific figures In addition to detailed coverage of specific instruments the book discusses analytical laboratory practices instrument maintenance statistics and real world lab experiments with previous student results Each analytical method section includes extensive sample preparation information rather than a simple stand alone chapter offering generic discussions not connected to specific methods This book conveniently organizes content by analyte class inorganic and organic in a way that is intuitive to a student and aligned with relevant courses Ancillaries including mp4 videos instructor PowerPoint slides and animations are included on a companion website Written by an experienced professor and tested and refined over years in his courses since 2008 Essential Methods of Instrumental Analysis includes information on sample topics such as Proper laboratory protocols for analytical instrumentation covering chemical reagents glassware calibration techniques and figures of merit Optical physics covering the interaction of electromagnetic radiation with instrument components and sample molecules relaxation processes reflection diffraction dispersion and refraction Flame atomic absorption and flame emission spectrometry covering optical radiation sources mirrors choppers burner heads and doppler broadening Gas and liquid chromatography covering gaseous liquid soil sediment and biological samples analyte recovery chromatography theory injectors columns and ovens common detectors and mass spectrometers Focusing on contrasts and comparisons across multiple types of instruments in a way distinct from similar texts Essential Methods of Instrumental Analysis is an essential textbook for students in advanced undergraduate courses in related programs of study

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*Chemistry* James E. Brady, John R. Holum, 1995-12-29 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping readers develop skills in analytical thinking and problems solving

General Chemistry Hunt, 1976-06-07

Quantitative Chemical Analysis Daniel C. Harris, Chuck Lucy, 2015-05-29

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web dec 14 2003 at the time appointed i will return unto thee according to the time of life and sarah shall have a son iv purpose to motivate the hearers to trust god to

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web ostensibly sarah s laughter expresses astonishment and joy just like the laughter of avraham in the preceding chapter 17



17 when he is told after his circumcision and

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web 15 sarah was afraid so she lied and said i did not laugh but he said yes you did laugh i m bothered by the last verse why does the text recount that sarah denied that