

Name: _____ Period: _____ Date: _____

Solubility and Concentration Pretest

1. – 2. I mix up a solution of some Kool Aid by using the mix, sugar and water. What is the *solute* and what is the *solvent*?

3. – 5. Define these terms:

Saturated:

Unsaturated:

Supersaturated:

6. I add some more sugar to the Kool Aid from 1. – 2. and it dissolves. Which of the above three terms best describes the solution? Explain.

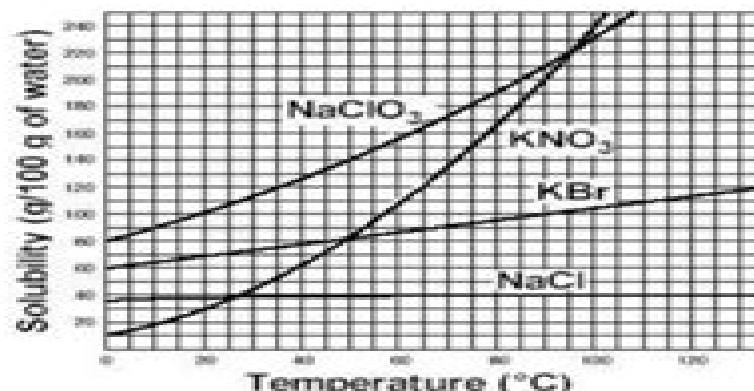
7. _____ How many grams of KBr are needed to make a saturated solution in 100 g of water at 30 °C?

8. _____ In terms of saturation, how would you classify a KBr solution with a concentration of 90 g / 100 g of water at 50 °C?

9. _____ If we raised the temperature to 100 °C, what would its saturation be?

10. _____ At what temperature is the solubility of KBr the same as that of KNO₃?

11. _____ Which is more soluble in water, NaClO₃ or KBr? How can you tell?



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N Colangelo



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