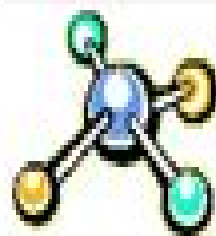


Name: _____ Date: _____ Class period: _____



Unit: Chemical Bonding **Answer Key** "Covalent Bonding" - WS #3

Directions: Please answer each question using your notes from today's class. ☺

1. What is the difference between a molecular formula, structural formula and an electron dot formula? Give an example of each.

A molecular formula gives the chemical make-up of one molecule. It does not show atom-to-atom connectivity. Example: C_2H_2

A structural formula uses lines instead of dots to show shared pairs of electrons.

Example: $H-H$

An electron dot formula uses dots instead of lines to show shared pairs of electrons.

Example: $H:H$

2. What is the difference between single, double, and triple covalent bonds? How many shared electrons are in each?

The difference between the types of bonds is: a single bond is two shared electrons, a

Concepts For Today Answer Key

National Academy of Sports Medicine

A red circular graphic with a gradient, appearing as a partial circle or a thick arrow pointing to the right, located to the right of the text 'National Academy of Sports Medicine'.

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