



Name: \_\_\_\_\_  
Date: \_\_\_\_\_

# COUNTING ATOMS

## WORKSHEET

- Chemical formulas can have three components.
  - The \_\_\_\_\_ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
  - The \_\_\_\_\_ which are numbers that are found on the lower \_\_\_\_\_ - hand side of each element symbol. This tells you the number of \_\_\_\_\_ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only \_\_\_\_\_ atom of this element in the molecule.
  - The \_\_\_\_\_ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be \_\_\_\_\_ by the \_\_\_\_\_ on the outside of the bracket found on the lower \_\_\_\_\_ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH	
Element	# of Atoms
Total	

4) Li <sub>2</sub> SO <sub>4</sub>	
Element	# of Atoms
Total	

7) H <sub>2</sub> COCH <sub>3</sub>	
Element	# of Atoms
Total	

10) NH <sub>4</sub> C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	
Element	# of Atoms
Total	

13) Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	
Element	# of Atoms
Total	

2) HNO <sub>3</sub>	
Element	# of Atoms
Total	

5) NH <sub>4</sub> Cl	
Element	# of Atoms
Total	

8) Mg(OH) <sub>2</sub>	
Element	# of Atoms
Total	

11) NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	
Element	# of Atoms
Total	

14) Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	
Element	# of Atoms
Total	

3) U <sub>2</sub> O	
Element	# of Atoms
Total	

6) CaClO <sub>4</sub>	
Element	# of Atoms
Total	

9) Al(OH) <sub>3</sub>	
Element	# of Atoms
Total	

12) (NH <sub>4</sub> ) <sub>3</sub> PO <sub>4</sub>	
Element	# of Atoms
Total	

15) Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	
Element	# of Atoms
Total	

# Counting Atoms In Compounds Answer

**August Dupré**



## Counting Atoms In Compounds Answer:

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