

Carbon and its compounds class 10 questions with answers

1. Define allotropy. Give two examples of allotropes of carbon.

Answer:

Allotropy is the existence of an element in different forms with the same physical state but different chemical properties.

Examples of carbon allotropes are diamond and graphite.

2. Explain the term 'homologous series' with an example.

Answer:

A homologous series is a series of organic compounds having the same functional group and a similar structure, with each successive member differing by the addition of CH_2 .

Example: The alkanes (saturated hydrocarbons) form a homologous series: methane (CH_4), ethane (C_2H_6), propane (C_3H_8) and so on.

3. What is the difference between a saturated and an unsaturated hydrocarbon? Give examples.

Answer:

Saturated hydrocarbons have only single bonds between carbon atoms.

Example: Alkanes like methane (CH_4).

Unsaturated hydrocarbons have at least one double or triple bond between carbon atoms. Examples: Alkenes like ethene (C_2H_4) and alkynes like ethyne (C_2H_2).

4. Explain the process of polymerization with an example.

Answer:

Polymerization is the process of combining small molecules (monomers) to form large molecules (polymers).

Example: The polymerization of ethene (C_2H_4) produces polyethylene, a common plastic.

5. Differentiate between structural isomers and stereoisomers.

Carbon And Hydrocarbons Answers

Rudolph Max Goepp



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