



Name: \_\_\_\_\_  
Date: \_\_\_\_\_

# COUNTING ATOMS

## WORKSHEET

- Chemical formulas can have three components.
  - The \_\_\_\_\_ which are represented by a single uppercase letter, or represented by an uppercase letter followed by a lowercase letter. This tells you the types of elements in the compound.
  - The \_\_\_\_\_ which are numbers that are found on the lower \_\_\_\_\_ - hand side of each element symbol. This tells you the number of \_\_\_\_\_ of this element in the molecule. If an element symbol has no subscript next to it, then this indicates that there is only \_\_\_\_\_ atom of this element in the molecule.
  - The \_\_\_\_\_ which surround some groups of atoms indicate that the numbers of all of the atoms inside the bracket need to be \_\_\_\_\_ by the \_\_\_\_\_ on the outside of the bracket found on the lower \_\_\_\_\_ - hand side of that bracket.
- Record the number of each atom in each molecule, then record the total number of atoms in the molecule:

1) NaOH	
Element	# of Atoms
Total	

4) Li <sub>2</sub> SO <sub>4</sub>	
Element	# of Atoms
Total	

7) H <sub>2</sub> COCH <sub>2</sub>	
Element	# of Atoms
Total	

10) NH <sub>4</sub> C <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	
Element	# of Atoms
Total	

13) Ca <sub>3</sub> (PO <sub>4</sub> ) <sub>2</sub>	
Element	# of Atoms
Total	

2) HNO <sub>3</sub>	
Element	# of Atoms
Total	

5) NH <sub>4</sub> Cl	
Element	# of Atoms
Total	

8) Mg(OH) <sub>2</sub>	
Element	# of Atoms
Total	

11) NaC <sub>2</sub> H <sub>3</sub> O <sub>2</sub>	
Element	# of Atoms
Total	

14) Fe <sub>2</sub> (CO <sub>3</sub> ) <sub>3</sub>	
Element	# of Atoms
Total	

3) U <sub>2</sub> O	
Element	# of Atoms
Total	

6) CaClO <sub>3</sub>	
Element	# of Atoms
Total	

9) Al(OH) <sub>3</sub>	
Element	# of Atoms
Total	

12) (NH <sub>4</sub> ) <sub>3</sub> PO <sub>4</sub>	
Element	# of Atoms
Total	

15) Al <sub>2</sub> (SO <sub>4</sub> ) <sub>3</sub>	
Element	# of Atoms
Total	

# Counting Atoms Test

**SA Adler**



## Counting Atoms Test:

SAT Subject Test Chemistry Joseph A. Mascetta, Mark Kernion, 2020-12-01 The updated edition of Barron's SAT Subject Test Chemistry includes A full length diagnostic test with explained answers Four practice tests that reflect the actual SAT Subject Test Chemistry All questions answered and explained Detailed reviews covering all test topics Appendixes which include the Periodic Table important equation constant and data tables and a glossary of chemistry terms Both teachers and test taking students have praised earlier editions of this manual for its wealth of well organized detail Subject reviewed include the basics matter energy scientific method and measurements atomic structure and the periodic table bonding chemical formulas gases and laws stoichiometry liquids solids and phase changes chemical reactions and thermochemistry chemical reactions chemical equilibrium acids bases and salts oxidation reduction carbon and organic chemistry and the laboratory ONLINE PRACTICE TESTS Students who purchase this book or package will also get access to two additional full length online SAT Chemistry subject tests with all questions answered and explained SAT Subject Test: Chemistry with Online Tests Joseph A. Mascetta, Mark Kernion, 2018-09-01 Always study with the most up to date prep Look for SAT Subject Test Chemistry ISBN 9781506263120 on sale December 01 2020 Publisher's Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entitles included with the product

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structure of physical science by emphasizing science as a search for truth rather than an accumulation of facts It develops the subject through concrete examples such as inclined planes and levers speeds and displacements progressing to consideration of forces and the concept of inertia and the idea of energy Similarly a study of observable chemical reactions advances to the ways in which atoms combine separate and displace one another and how observable masses of reactants and products illustrate how the atoms are combining The fundamental ideas are applied to astronomy optics geology music and the chemical compounds of life processes The wide variety of end of chapter problems and multiple choice questions reinforce comprehension of each topic

**Chemistry** Neil D. Jespersen, Alison Hyslop, 2021-12-03 Chemistry The Molecular Nature of Matter 8th Edition continues to focus on the intimate relationship that exists between structure at the atomic molecular level and the observable macroscopic properties of matter Key revisions in this edition focus on three areas The deliberate inclusion of more updated real world examples that relate common real world student experiences to the science of chemistry Simultaneously examples and questions have been updated to align them with career concepts relevant to the environmental engineering biological pharmaceutical and medical sciences Providing students with transferable skills with a focus on integrating metacognition and three dimensional learning into the text When students know what they know they are better able to learn and incorporate the material Providing a total solution through New WileyPLUS by fully integrating the enhanced etext with online assessment answer specific responses and additional practice resources The 8th edition continues to emphasize the importance of applying concepts to problem solving to achieve high level learning and increase retention of chemistry knowledge Problems are arranged in an intuitive confidence building order

**LISP** Patrick Henry Winston, 1984 This third edition is a revised and expanded version of Winston and Horn's best selling introduction to the LISP programming language and to LISP based applications many of which are possible as a result of advances in Artificial Intelligence technology

**Nondestructive Testing**, 1950

**A Learning Program for Chemistry** Peter Demmin, 1993-11

**Virtual Chemlab** Brian F. Woodfield, Matthew C. Asplund, Steven Haderlie, 2006 Contains a full virtual lab environment as well as the pre arranged labs that are referenced in the workbook and at the end of the chapter in the textbook Virtual ChemLab can be run directly from the CD or installed on the student's computer

**Object-oriented Common LISP** Stephen Slade, 1998 LISP was developed in the late 1950s as a language for manipulating symbols This book presents the Common LISP programming language which is a version of LISP and details its range of application including data structures computer systems and compiler design It provides extensive examples of LISP programs in a variety of areas such as text formatting and spelling correction

**Scientific American**, 1913

Leak Testing Robert Charles McMaster, 1982

*Standard Specifications for Transportation Materials and Methods of Sampling and Testing: Tests* American Association of State Highway and Transportation Officials, 1995

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Officials,1986      *Journal of Chemical Education* ,1929      **Steam** ,1914      *Practical Chemistry* Lyman Churchill  
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