12. unshared electron pair

Covalent Bonding Quiz (24 Points)

Name:

H. a covalent bond in which three pairs of electrons

are shared by two bonded atom

Section I - Basic Concepts & Vocabulary (12 points - 1 pt. each)

Instructions: For Items 1-4, circle the correct response.

1.	A covalent bond is formed by				electrons between two nonmetal atom
	a)	freezing	c)	sharing	

b) transferring d) melting

2. A group of nonmetal atoms, joined by covalent bonds is defined as a(an)

a) molecule c) metal b) salt crystal d) octet

 A structural formula uses Lewis dot structures to show how atoms are bonded, including the number & types of bonds, in a covalent molecule? (Yes or No)

4. Do atoms, except hydrogen, in covalent bonding follow the octet rule? (Yes or No)

Instructions: Match the word on the left with the definition on the right by writing the letter on the line next to the word that is a "definition match" for that word.

5. molecular formula	 A. a covalent bond in which only one pair of electrons is shared by two bonded atoms 			
	6. nonpolar covalent bond	B. chemical		
	formula that shows the arrangement of			
	atoms in molecules and pol-	atoms in molecules and polyatomic ions		
7. triple covalent bond	C. a covalent bond b	etween two atoms of		
	different electronegativities in which the bonding electrons are not shared equally			
8. double covalent bond	D. a covalent bond f	ormed by the equal		
	sharing of bonding electron	s by two atoms		
9. polar covalent bond	 E. a covalent bond in which are shared by two bonded a 			
10, single covalent bond	F. two valence electrons tha	at are not involved in		
	bonding at a bond site in a	covalent molecule		
11. structural formula	G. indicates the exact numb	er of atoms in a		
	covalent compound, but do	es NOT show the		
	arrangement of the atoms r	or types of bonds		

Covalent Bonding Quiz With Answers Student Edgenuity

DJ Losen

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