Justin Silva

1) Indicate the oxidation number of carbon and sulfur in the following compounds.

a.
$$CO = (C = +2)$$

b.
$$CO2 = (C = +4)$$

c.
$$Na2CO3 = (C = +4)$$

d.
$$Na2C2O4 = (C = +3)$$

e.
$$CH4 = (C = -4)$$

f.
$$H2CO = (C = 0)$$

g.
$$SO2 = (S = +4)$$

$$b. SO3 = (S = +6)$$

i.
$$Na2SO4 = (S = + 6)$$

$$j. Na2SO3 = (S = +4)$$

$$k. Na2S2O3 = (S = +2)$$

1.
$$Na2S4O6 = (S = +5/2 = +2.5)$$

$$m. SC12 = (S = +2)$$

$$n. Na2S2 = (S = -1)$$

o.
$$SOC12 = (S = +4)$$

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