

Program: B.Tech
Course Code & Title: 21CVB101J/Chemistry
Year & Sem: I/II
Date: 29-03-2023
Time: 12.30-1.30 pm
Max. Marks: 30 Marks
Answer ALL the MCQs Part-A (10*1=10 Marks)

- Which of the following is not a state function
 a) ΔG b) ΔE c) W d) H
- Internal energy does not include
 a) Nuclear energy b) Vibrational energy c) Rotational energy d) Energy of gravitational pull
- Gibbs energy change ΔG is related to equilibrium constant K as:
 a) $\Delta G^\circ = -RT \ln k$ b) $\Delta G^\circ = RT \ln k$ c) $\ln k = -RT/\Delta G^\circ$ d) $\ln k = \Delta G^\circ/RT$
- The conductivity of electrolytic conductors is due to _____
 a) Flow of free mobile electrons b) Movement of ions c) Either movement of electrons or ions d) Cannot be said
- The process of deterioration of a metal due to unwanted chemical or electrochemical interaction of the metal with its environment is called _____
 a) Electrolysis b) Electrodialysis c) Corrosion d) Deposition
- Chemical used in salt bridge is
 a) KOH b) KCl c) KNO_3 d) KBr
- If a compound has 3 chiral carbons What is the number of optically active isomers?
 a) 9 b) 3 c) 4 d) 8
- The potential energy of n-butane is minimum for
 a) Skew conformations b) Staggered conformations c) Eclipsed conformations
 d) Gauche
- Optical isomerism is a type of _____
 a) metamerism b) stereoisomerism c) geometrical isomerism d) tautomerism

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Judd E. Hollander



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