

## Unit 5 – Counting Particles - Objectives

|                                                                                                                                                                                                      |  |
|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|--|
| <p>Review Concepts</p> <ol style="list-style-type: none"> <li>a. Types of substances</li> <li>b. Chemical formulas of substances (U 4)</li> </ol>                                                    |  |
| <p>1. State evidence for Avogadro's Hypothesis. Use Avogadro's Hypothesis and experimental data to determine the relative mass of molecules.</p>                                                     |  |
| <p>2. Use experimental data to determine the relative mass of two objects.</p>                                                                                                                       |  |
| <p>3. Use experimental data to determine the number of items in a sample without actually counting them.</p>                                                                                         |  |
| <p>4. Given the chemical formula of a substance, determine the molar mass.</p>                                                                                                                       |  |
| <p>5. Given the mass of a substance, determine</p> <ol style="list-style-type: none"> <li>a. the number of moles of the sample</li> <li>b. the number of atoms or molecules in the sample</li> </ol> |  |

# Chemistry Unit 5 Counting Particles Key

**Sir Norman Lockyer**



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