

## Chemistry – Unit 6 Worksheet 1

We have observed evidence that atoms of different elements are different, the most noticeable being their predictable characteristic chemical behaviour. Atoms of different elements react in fixed, repeatable, predictable ways. However, when the reactions started, scientists needed justification for attributing to their components the same chemical behaviour as atoms. We will now examine the patterns that reveal the structure of atoms that allows them to react in predictable ways for different elements.

1. Write the formula and draw the particle diagram for each compound. The ratio of atoms in each compound is given.

Atomic formula and compound	12 hydrogen 12 nitrogen	12 hydrogen 12 nitrogen	12 hydrogen 12 carbon	1 hydrogen 1 carbon
	$\text{CH}_4$	$\text{C}_2\text{H}_6$	$\text{C}_2\text{H}_6$	$\text{CH}_4$

particle



diagram

Atomic formula and compound	12 carbon 12 nitrogen	12 hydrogen 12 nitrogen	12 carbon 1 carbon
	$\text{C}_2\text{H}_6$	$\text{C}_2\text{H}_6$	$\text{C}_2\text{H}_6$

particle



diagram

Atomic formula and compound	12 hydrogen 12 carbon	12 hydrogen 12 carbon	1 hydrogen 1 carbon	1 carbon 1 hydrogen
	$\text{CH}_4$	$\text{CH}_4$	$\text{CH}_4$	$\text{CH}_4$

particle



diagram

# Chemistry Unit 6 Test

**Society of Chemical Industry (Great Britain)**



## **Chemistry Unit 6 Test:**

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