

### Information: Changes of Matter: Temperature Dependence

Scientists have learned that certain particles that a large mass of has sufficient mass and volume to be able to be used as a source of energy. Scientists also have learned that a certain amount of energy is required to change the state of matter. The first law of thermodynamics states that the total energy of a system is constant.

Return to Table 18.1. Return to Table 18.2. Return to Table 18.3. Return to Table 18.4. Return to Table 18.5.

### Critical Thinking Questions

1. Explain the first law of thermodynamics in your own words.

See Table 18.1  
Return to Table 18.1

See Table 18.2  
Return to Table 18.2

See Table 18.3  
Return to Table 18.3

### Information: More Than One Possible Change

Many materials can be used to make a change in the state of matter. For example, water can be used to make a change in the state of matter. The first law of thermodynamics states that the total energy of a system is constant. The second law of thermodynamics states that the total entropy of a system is constant.

### Critical Thinking Questions

2. Explain the second law of thermodynamics in your own words. How does it relate to the first law of thermodynamics? How does it relate to the third law of thermodynamics?

See Table 18.1  
Return to Table 18.1

See Table 18.2  
Return to Table 18.2

See Table 18.3  
Return to Table 18.3

See Table 18.4  
Return to Table 18.4

3. Explain the third law of thermodynamics in your own words. How does it relate to the first law of thermodynamics? How does it relate to the second law of thermodynamics?

See Table 18.1  
Return to Table 18.1

See Table 18.2  
Return to Table 18.2

See Table 18.3  
Return to Table 18.3

See Table 18.4  
Return to Table 18.4

### Information: First Law of Thermodynamics: Energy Conservation

The first law of thermodynamics states that the total energy of a system is constant. The second law of thermodynamics states that the total entropy of a system is constant. The third law of thermodynamics states that the total volume of a system is constant.

See Table 18.1  
Return to Table 18.1

See Table 18.2  
Return to Table 18.2

See Table 18.3  
Return to Table 18.3

See Table 18.4  
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### Critical Thinking Questions

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**J Dewey**



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