

Name _____

Class _____

Date _____

Skills Worksheet

Concept Review**Section: Concentration and Molarity**

Complete each statement below by choosing a term from the following list.
Use each term only once.

concentration

moles

solute

solution

liter

parts per million

molality

molarity

ratios

1. _____ is the quantity of solute in a specific quantity of solvent or solution. One way of expressing concentration is to use the unit of measure called _____; it is the ratio of _____ amount (in moles) to _____ volume (in liters), or simply _____ of solute per L of solution. Table 1 in your textbook lists other concentration units or _____, such as _____ ($m = \text{mol solute/kg solvent}$) and _____ ($\text{ppm} = \text{mL solute/1 000 000.0 mL solution}$).

Solve the following problems and write your answer in the space provided.

2. What is the molarity of a hydrochloric acid solution that has a volume of 1500 mL and contains 441 g of HCl?
3. What is the molarity of a sugar solution that has a volume of 0.500 L and contains 17.1 g of ordinary sugar ($\text{C}_{12}\text{H}_{22}\text{O}_{11}$)?

Concept Review Section Concentration And Molarity

Anthony S. Fauci



Concept Review Section Concentration And Molarity:

Holt Chemistry Holt Rinehart & Winston,2003-01-24 Ebook: Chemistry: The Molecular Nature of Matter and Change Silberberg,2015-01-16 Ebook Chemistry The Molecular Nature of Matter and Change *Chemistry*³ Andrew Burrows,Andrew Parsons,Gwen Pilling,Gareth Price,2013-03-21 New to this Edition **Chemistry3** Andrew Burrows,John Holman,Andrew Parsons,Gwen Pilling,Gareth Price,2017 Chemistry3 establishes the fundamental principles of all three strands of chemistry organic inorganic and physical By building on what students have learned at school using carefully worded explanations annotated diagrams and worked examples it presents an approachable introduction to chemistry and its relevance to everyday life **CHEMISTRY** SILBERBERG,2003 *Holt Chemistry* Ralph Thomas Myers,2004 General Chemistry Raymond E. Davis,Kenneth W. Whitten,1996 Are you looking for the key to success in your chemistry class In CHEMISTRY you will find a strong molecular reasoning focus problem solving exercises and an innovative online homework management system that will prepare you for any challenge you might encounter The textbook is filled with learning aids that will help you master concepts of the course **Complete Preparation for the AHPAT, 1999** Williams & Wilkins Review,1998-04 Here is the most accurate and complete resource designed for students interested in applying for admission to health fields such as medical technology and medical therapy Reviews cover science knowledge verbal and quantitative ability and reading comprehension Students will value the topic outlines for the AHPAT exam the in depth allied health glossary and the full length sample test complete with solutions **VETs, Complete Preparation for the Veterinary Entrance Tests** Aftab S. Hassan,1998 *VETs, Complete Preparation for the Veterinary Entrance Tests* ,2000 **A Complete Preparation for the MCAT** Aftab S. Hassan,James L. Flowers,1992 This guide for MCAT preparation applies the principles of active and problem based learning to an updated review of content and skills with models for enhanced problem solving and critical thinking abilities There are details on setting up a self managed study programme with guidelines for time management and stress management All areas tested on the exam are covered verbal reasoning physical science writing sample biological sciences with practice questions to chart progress **Allied Health Professions Admissions Test** Aftab Hassan,Betz Guide,Aftab S Hassan, Ph. D.,Williams & Wilkins Lippencott,1995 This guide is aimed at students interested in allied health careers and covers science knowledge verbal and quantitative ability and reading comprehension based on the methods of active learning The text guides students through a system of self managed self paced study related to the AHAT Each section offers AHPAT style practice questions and there is also a model examination with answers and explanations The appendix includes a glossary allied health careers **PCAT** Aftab S. Hassan,Williams & Wilkins Review,Betz,1998 Based on Guidelines from the American Association of Colleges of Pharmacy this volume includes a skills based review of the latest content and a full length practice exam with solutions Readers will acquire the skills and the secrets that will lead to improved success including a 10 point prescription for conquering the exam *Chemistry* Martin Stuart Silberberg,1996

This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change. The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives.

Analytical Chemistry Gary D. Christian, 1986-01-17 This is a practical approach to quantitative analytical chemistry covering all areas of modern quantitative analysis taught in a standard first course in quantitative analysis. Includes experiments in each method. This edition includes coverage of electronic balance and propagation of error. Equilibria are introduced in terms of Gibbs free energy buffers and calculations are presented in terms of proton acceptor/donor. Experiments are now all at the back of the book. SI units are emphasized throughout. Numerous applications to the life sciences.

Introductory Chemistry Steve Russo, Mike Silver, 2002 Steve Russo and Mike Silver turn chemistry into a memorable story that engages readers and provides the context they need to understand and remember core concepts. The book builds interesting applications and well designed illustrations into the narrative to get and hold attention then builds confidence with integrated active learning activities. Readers make the connections between concepts and the problem solving techniques they need to master as they read. The new edition strengthens this conceptual approach and presents additional quantitative techniques in key areas. Readers will find enhanced support for quantitative problem solving and more challenging questions at the end of each chapter in addition to the wealth of technology based support on The Chemistry Place™ Special Edition and on The Chemistry of Life CD ROM. For college instructors and students.

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Fundamentals of Chemistry David E. Goldberg, 2001

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