Equilibrium Systems 3H2(9) + N2(9) = 2NH3(9) Kp = 2.25 × 10- at 600 °C

Stress 1: adding / removing equilibrium component (remove H, shift (-)

Stress 2: Change overall pressure (increase pressure by decreasing vol., shift -)

Stress 3: Changing temp. of System (increase temp, shift -)

Stress 4: (hanging a compound related to the reaction compo

Eq. doesn't change when adding/removing 1 or 5

Q = equilibrium expression

a=k equilibrium OLK - shift toward products 01k + shift toward reactants

Concept Review Equilibrium Systems And Stress

Clifford Lane

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