

## Quiz & Worksheet - Double Displacement Reaction

<http://study.com/academy/practice/quiz-worksheet-double-displacement-reaction.html>

1. What type of reaction is the following?



☐ combustion

☐ gas formation

☐ precipitation

☐ neutralization

2. What are the ions and the correct charges for the following compound:



☐  $\text{Pb}^{2+}$   $\text{Cl}^{2-}$

☐  $\text{Pb}^{2-}$   $\text{Cl}^{2-}$

☐  $\text{Pb}^{+}$   $\text{Cl}^{2-}$

☐  $\text{Pb}^{2+}$   $\text{Cl}^{-}$

3. What are the correct products for the following reaction? (also include the correct coefficient for a balanced reaction)



☐  $\text{Ag}_2\text{SO}_4 + \text{Na}_2\text{NO}_3$

☐  $\text{Ag}_2\text{SO}_4 + \text{NaNO}_3$

☐  $\text{Ag}_2\text{SO}_4 + 2\text{NaNO}_3$

☐  $\text{AgSO}_4 + \text{NaNO}_3$

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# Double Replacement Reaction And Solubility Lab Answer Key

**Valentine Korah**



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think about a phenomenon in a specific way d To apply personal experience in answering a questions Pref      **Bibliography**  
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