

# Concept Review

## Section 8-1: Covalent Bonds

Answer the following items in the space provided.

1. What does a molecular bond-forming reaction look like?
2. Why is the  $H_2$  molecule more stable than two separate hydrogen atoms?
3. Explain why the stability difference in  $H_2$  does not show that bond-forming is more energetically favorable.
4. What does a molecular bond-forming reaction look like?
5. What is enthalpy change?
6. Describe potential energy changes when two atoms form a diatomic molecule. Explain why the bond-forming reaction is energetically favorable.
7. In terms of energy, why is the  $H_2$  bond stable?
8. Why is bond length an average value? How is bond length?
9. Describe the physical properties of molecules that are formed by covalent bonds.

Complete each statement below by choosing a term from the following list. Some may be used more than once.

- |                      |                      |                         |                      |
|----------------------|----------------------|-------------------------|----------------------|
| covalent bond        | electron sharing     | electron                | single covalent bond |
| double covalent bond | triple covalent bond | quadruple covalent bond |                      |
1. Bond \_\_\_\_\_ is the force that holds two atoms together. It is the energy that holds two atoms together.
  2. Bond \_\_\_\_\_ is the force that holds two atoms together. It is the energy that holds two atoms together.
  3. The energy of an atom is the energy that holds two atoms together. It is the energy that holds two atoms together.
  4. In \_\_\_\_\_, two atoms share two electrons. This is the energy that holds two atoms together.
  5. In \_\_\_\_\_, two atoms share four electrons. This is the energy that holds two atoms together.

# Concept Review Section Covalent Bonds 6 Answers

**Daniel D. Chiras**



## Concept Review Section Covalent Bonds 6 Answers:

**Concepts in Biology' 2007 Ed.2007 Edition** , Science Tutor, Grades 6 - 8 Raham,2008-09-02 Connect students in grades 6 and up with science using Science Tutor Earth and Space This effective 48 page resource provides additional concept reinforcement for students who struggle in earth and space science Each lesson in this book contains an Absorb section to instruct and simplify concepts and an Apply section to help students grasp concepts on their own The book covers topics such as the layers of the earth types of rock how rock is formed weather the phases of the moon and Earth s place in the solar system It also highlights key terms in the text and includes a recap of the metric system The book supports National Science Education Standards *Basic Concepts of Chemistry, Study Guide* Leo J. Malone,2000 The text s three main goals are to introduce chemistry as a living relevant science to encourage learning and critical thinking and to help readers overcome the math difficulties that impede their progress in chemistry Designed to help readers master the principles of general chemistry As a prep book it promotes active involvement with the material There are special features throughout that reinforce concepts and help to develop strong problem solving and study skills Updated to Include an Interactive Learning Ware problems CD containing several of the chapter ending problems from the book in an interactive tutorial with feedback to help readers set up and solve problems *Study Guide to accompany Basic Concepts of Chemistry, 7th Edition* Leo J. Malone,2003-02-20 Work more effectively and gauge your progress along the way This Study Guide that is designed to accompany Malone s Chemistry 7th Edition includes chapter summaries new terms self tests answers to self tests and solutions to selected problems This easy to read introduction presents chemistry as a living relevant science Chemistry 7th Edition encourages critical thinking and helps readers overcome the math difficulties that often prevent them from developing a full understanding of the subject This new seventh edition builds on its core strengths of pedagogy driving the connections between ideas mathematics in context not just an appendix and an extensive problem solving emphasis with an updated design and more molecular art In addition the seventh edition expands its applications and online options One of the briefest books in the market it still provides sufficient depth for the basic concepts of chemistry Fundamentals of Microbiology Jeffrey C. Pommerville,2014-12 Ideal for health science and nursing students Fundamentals of Microbiology Body Systems Edition Third Edition retains the engaging student friendly style and active learning approach for which award winning author and educator Jeffrey Pommerville is known Highly suitable for non science majors the fully revised and updated third edition of this bestselling text contains new pedagogical elements and an established learning design format that improves comprehension and retention and makes learning more enjoyable Unlike other texts in the field Fundamentals of Microbiology Body Systems Edition takes a global perspective on microbiology and infectious disease and supports students in self evaluation and concept absorption Furthermore it includes real life examples to help students understand the significance of a concept and its application in today s world whether to their local community or beyond New information

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