200

4 de de

1000









40.76

concentration of reactants. The greatest the concentration of reactants from the property the mark frequently they collide with one

this equation in words from does this equation relate to lineathing?

. For imagement winners, see Approach A.

CHAPTER 2: The Chemical Context of Life

2 Chapter Review



 Go to Mastaring Biology for Assignments, the situal, the Study Area, and Dynamic Study Modules.

SUMMARY OF KEY CONCEPTS

 To review key terms, go to the Vocabulary Self-Quis in the Mastering Biology effect or Study Jens, or go to good physics.

CONCRET 2.1

Matter consists of chemical elements in pure form and in combinations called compounds (co. 29-30)

- Elementals cannot be broken down chemically to other substances. A composant contains the contai
- Compare an alement and a compound

COMOLING 202

An element's properties depend on the structure of its atoms (pp. 30-36)

 Act actions, the smallest unit of an element, has the following components:



- An observationly neutral atoms has equal numbers of electronic and proteins, the number of proteins determined the atoms number. The atomic means is moreover in distincts and it roughly equal to the mass number, the sum of protein plan tourisons. Bedopper of an element differ from such other in neutron number and therefore man. Unstable hootopes give off position and energy as about a discount of the con-
- In an atom, electrons occupy specific electron shells; the electrons is a chell bare a characteristic energy level. Electron distribution in a chell-determine the chemical behavior of an atom. An atom that has an incomplete outer shell, the valence shell, is multiple.
- Elloy throug exist in orbitals, there dimensional spaces with specific shapes that are components of electron shells.

DRAWN RT. Draw the electron distribution diagrams for moon (LyAe) and argon (LyAE). Use these diagrams to explain with these elements are charactery consistent and charactery consistent.

сомоция 200

The formation and function of molecules and ionic compounds depend on chemical bonding between atoms (pp. 38-40)

 Chemical bossels form when atoms interact and complete their valence shells. Covalent bossels form others pain of electrons are shared.

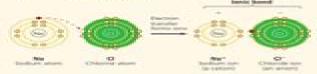


Q unit one. The Chemistry of Life

Meliocuties notestated of two or more considerably bounded attents. The attentions of an attent the tile electrons of a consultent bound in its electron organization. If books attents are the same, they have the same electron against my and share a monopoliar correlated bound. The transmission of a poliar considerat bound are publied closes in the more

elisoctroninggetiver urborn, much as other onlygons in HACC.

An Laws Scotter whom an attention or mandercalar gains or frozen are effections and florest are effections and florecomes charged. An locate bound to the attraction between these approximates because it was not been provided by the partial florest process.



- Would internetions scientisely the drapps of large molecules, and help molecules adhere to each other. A bydrogen board to an atmaction between a hydrogen atom corrying a partial position change (b +) and an electrologistive alone carrying a partial regulary change (b +). Non ther Windo internetions occur between transiently positive and negative regions of molecules.
- A mechanist's shape in determined by the positions of its atoms' referre certains. Complete bounds provide in hybrid extension, which are responsible for the shapes of BaO, GE, and many more complete biological encolorates. Molecular shape is among the backs for the soccognition of one biological molecule by another.
- In forms of electron sharing between atoms, compare rongosiar socient bonds, polar covelent bonds, and the formation of sons.

CONCLET 2.4

Chemical reactions make and break chemical bonds

- Chemical resettions change reachasts into products their conserving matter. All chemical reactions are importable reversible.
 Chemical equilibrium is reached offer the forward and reverse resultion tests are equal.
- Michael vacculal happens to the concentration of products of more reactions; selected to a reaction that was not informed equilibrium? Assessment this addition of the equilibrium?

TEST YOUR UNDERSTANDING

 For more multiple-choice questions, go to the Practice Sect in the Mactering Biology effect or Study Area, or go to poolgining/life.

Levels 1-2: Remembering/Understanding

 Consequenced width ¹⁶T, the randome time invelope ¹⁶T has (A) a different attention tumber. (C) once there electrons.
 (E) once there electrons.

2. In the term from observe, the adjective from money that Levels

- GAy other elements in required on very small amounts. (B) the element can be used as a label to brace atomic through an organism's muriabolism.
- IC3 tibe element is very new on Earth.
- (D) the element enhances health but is not energial for the organism's long-term survival.
- The reactivity of an atom arises from: (A) the average distance of the outerment electron shell from the reactions.
- (B) the existency of sequenced electrons in the valence shell.
 (C) the some of the potential intergies of all the electron shells.
 (D) the potential energy of the tablence shell.

 8. White's statement is true of all atoms that are aniows?

Levels 5-6: Evaluating/Creating

- **DR. EMOLUTION CONNECTION** The percentages of sustantilly occurring elements making up the human body (see Table 2.1) are similar to the processings of those elements housed in other organisms. How could you account for this shealthing among organisms.
- B.B. SCREWING INCOMENT Permadehuma under Joy centertring chemical attract males by centertring chemical attract signate that spread it briough the six. A male basedowle of meters away can detect these molecules and Dy toward obserted. The memory organs



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L Cohen

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