

Name: \_\_\_\_\_

## Double Replacement Lab

Reminder – Goggles must be worn at all times in the lab!

### PRE-LAB DISCUSSION:

A PRECIPITATE is an insoluble substance formed by the reaction between chemicals in solution. A list of reactions will be given to you. The reactants are on the left side of the equation, and the products are on the right. The precipitate will be found on the product side. Keep in mind that an equation gives a condensed picture of what happens when two chemicals react.

### PURPOSE:

The purpose of the lab is to observe semi-micro scale precipitation reactions, and to use a solubility table to identify the insoluble product, when there is one.

### PROCEDURE:

Predict the products and balance the following double replacement reactions. Then, identify the precipitates using the table on page 2. **You may want to do your work step on a scratch piece of paper and attach to lab later.**

<u>Double Replacement Reactions</u>					<b>Precipitates (if any)</b>
A.	NaCl	+	AgNO <sub>3</sub>	→	
B.	KI	+	AgNO <sub>3</sub>	→	
C.	NaOH	+	AgNO <sub>3</sub>	→	
D.	Na <sub>3</sub> PO <sub>4</sub>	+	AgNO <sub>3</sub>	→	
E.	NaCl	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
F.	KI	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
G.	NaOH	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
H.	Na <sub>3</sub> PO <sub>4</sub>	+	Pb(NO <sub>3</sub> ) <sub>2</sub>	→	
I.	NaCl	+	CuSO <sub>4</sub>	→	
J.	KI	+	CuSO <sub>4</sub>	→	
K.	NaOH	+	CuSO <sub>4</sub>	→	
L.	Na <sub>3</sub> PO <sub>4</sub>	+	CuSO <sub>4</sub>	→	

# Double Replacement Lab Key

**Lei Huang**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the name bar.

## Double Replacement Lab Key:

**Regents Exams and Answers: Chemistry--Physical Setting Revised Edition** Barron's Educational Series, Albert Tarendash, 2021-01-05 Barron's Regents Exams and Answers Chemistry provides essential practice for students taking the Chemistry Regents including actual recently administered exams and thorough answer explanations for all questions This book features Eight actual administered Regents Chemistry exams so students can get familiar with the test Thorough explanations for all answers Self analysis charts to help identify strengths and weaknesses Test taking techniques and strategies A detailed outline of all major topics tested on this exam A glossary of important terms to know for test day **E3**

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