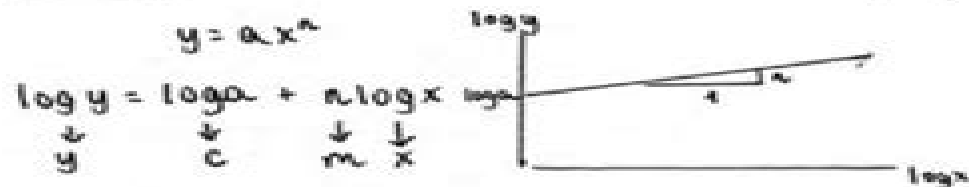


# STATISTICS

## 1) REGRESSION, CORRELATION AND HYPOTHESIS TESTING

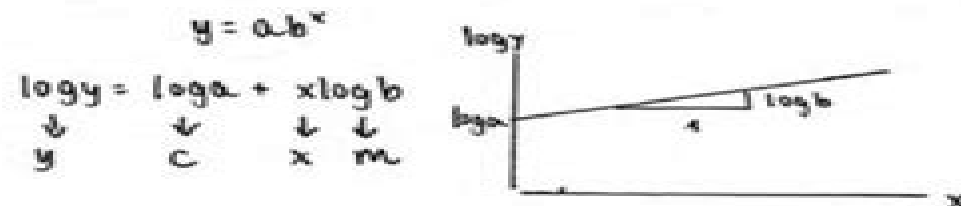
1.1) Exponential models → logarithms and coding used to examine trends in non-linear data.

$y = ax^n$  - power ( $a, n \rightarrow$  constant) → polynomial base unknown



graph:  $\log y$  against  $\log x$ , straight line with gradient  $n$  and vertical intercept  $\log a$ .

$y = ab^x$  - exponential ( $a, b \rightarrow$  constant) → exponential power unknown (exponent)



graph:  $\log y$  against  $x$ , straight line with gradient of  $\log b$  and vertical intercept  $\log a$ .

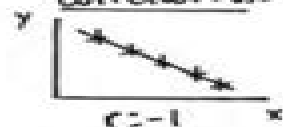
## 1.2 - measuring correlation

$r = \pm 1 \rightarrow$  straight line

The product moment correlation coefficient describes the linear correlation between two variables. It can take any value between -1 and 1.

→ PMCC measures the type and strength of, linear correlation ←

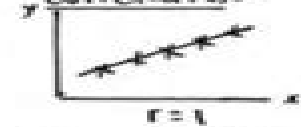
$r = -1$ , perfect negative correlation



← closer to 1 or -1; stronger the correlation →



$r = 1$ , perfect positive correlation



- $-0.2 < r < 0.2 \Rightarrow$  No significant correlation
- $-0.4 < r < -0.2$  /  $0.2 < r < 0.4 \Rightarrow$  weak correlation
- $-0.7 < r < -0.4$  /  $0.4 < r < 0.7 \Rightarrow$  moderate correlation
- $x < -0.7$  /  $x > 0.7 \Rightarrow$  strong correlation
- close to 1 or -1  $\Rightarrow$  very strong correlation

As [name of  $x$ ] increases,  
[name of  $y$ ] increases /  
decreases on average.

always say on average  
unless it is perfect.

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exothermic endothermic b how much heat will be absorbed or released if 10 g of NO is reacted with 10 g of  $\text{H}_2\text{O}$

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phase what is the heat of fusion formula  $h_n = h_{\text{fus}}$  what is the heat of vaporization formula  $h_n = h_{\text{vap}}$  what is on the x axis of the  
heating curve time

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option c is incorrect because 566 kJ mol is the  $\Delta h$  per mole of  $\text{O}_2$  and not for 2 moles even though options b and d are  
expressed in grams they are still correct because 56

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according to the following equation how much heat energy must be expelled by the body to rid the body of one pound of fat  
 $2\text{C(s)} + 3\text{O}_2\text{(g)} \rightarrow 2\text{CO}_2\text{(g)}$  7914 kJ is this endothermic or exothermic 2

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