

## Dilutions Worksheet

- 1) If I have 340 mL of a 1.5 M NaBr solution, what will the concentration be if I add 560 mL more water to it?
- 2) If I dilute 250 mL of 0.50 M lithium acetate solution to a volume of 750 mL, what will the concentration of this solution be?
- 3) If I leave 750 mL of 0.75 M sodium chloride solution uncovered on a windowsill and 150 mL of the solvent **evaporates**, what will the new concentration of the sodium chloride solution be?
- 4) To what volume would I need to add water to the evaporated solution in problem 3 to get a solution with a concentration of 0.50 M?
- 5) If I add 25 mL of water to 125 mL of a 0.15 M NaOH solution, what will the molarity of the diluted solution be?
- 6) If I add water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity of the diluted solution be?
- 7) How much 0.05 M HCl solution can be made by diluting 250 mL of 10 M HCl?

# Dilution Notes And Practice Answers

**JA Banks**



## **Dilution Notes And Practice Answers:**

## **Dilution Notes And Practice Answers** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Dilution Notes And Practice Answers**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we shall delve to the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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