

STOICHIOMETRY

Practice Problems

10 problems, 10 minutes each, for practice before the exam.

SECTION 12.1: CHEMICAL CALCULATIONS

1. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

2. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

3. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

4. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

5. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

6. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

7. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

8. A compound is found to consist of 40.0% oxygen, 40.0% carbon, and 20.0% hydrogen.

$$\text{Empirical formula} = \text{CH}_2\text{O}$$

Bonar Law Chapter 12 Practice Problems Stoichiometry Answers

VM Jensen



Bonar Law Chapter 12 Practice Problems Stoichiometry Answers:

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