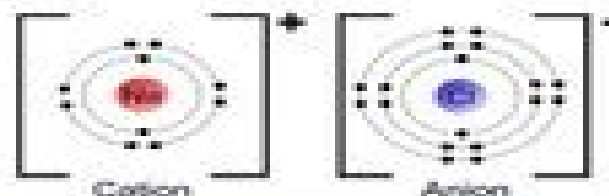
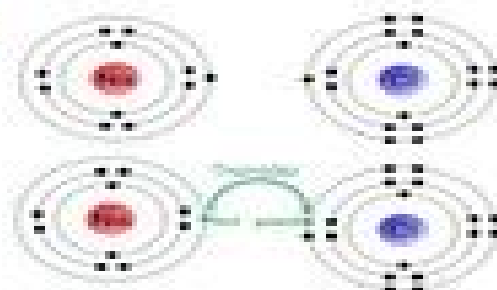


CHEMICAL BONDING

(Ionic Bonding)

The Valence

- The valence is the number of electrons lost or gained during the formation of ionic bond.
- Sodium atom loses an electron and becomes Na^+ ion, then the valence is 1.
- Chlorine atom gains an electron and becomes Cl^- ion then the valence is 1.



**Middle School Chemistry
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Fundamentals of Chemistry, *Fundamentals and Applications of Gas Phase Ion Chemistry* K.R. Jennings, 1999 This book draws together leading workers in the fields of mass spectrometry and gas phase ion chemistry providing a forum for the exchange of ideas and information It contains 14 lecture summaries and accounts of seven workshops with contributions from Europe and North America **Science Scope**, 2003 **15 PGT Chemistry Test Papers EMRS** Mocktime Publication, EMRS Exam Teachers PGT Chemistry Test Papers 15 Practice Papers Tier 1 Eklaya Model Residential Schools as per Official Exam Pattern and Syllabus *FUNDAMENTALS OF REACTION MECHANISMS IN ORGANIC CHEMISTRY* NARAIN, R. P., 2011-03 Written for the undergraduate and postgraduate students of chemistry this textbook presents comprehensive coverage of different types of reactions and their mechanisms The need for such a book has been felt for a very long time both by students and teachers The book discusses chemical kinetics structure and reactivity and reactive intermediates such as carbenes nitrenes and benzyne It also describes the mechanism of tautomerism and the concepts of aromaticity In addition the book elaborates the various reactions such as substitution free radical addition elimination and alkylation reactions Finally the text presents a detailed discussion on molecular rearrangements oximes and diazo compounds as well as the concepts of photochemistry **KEY FEATURES** Presents a number of examples to explain the mechanistic concepts Offers graphs and tables at various places to illustrate the key points Includes latest information on the subject **School**, 1918 *Chemistry Trends* Keith B. Hutton, 2024-12-15 Today's fast moving world of science will have far reaching effects on all of our lives Trends in Science is a series of essential readings for anyone who wants to know more about how his or her future will be affected as well the series provides accessible and stimulating material for high school and college students for researchers and librarians All titles in the series provide an introductory overview of the field in the last 100 years reviewing the past but also predicting the new developments of the future a detailed chronology of the most important milestones an index of key terms and concepts biographies of the most important scientists in each field and their role in shaping their particular branch of science a listing of important Websites a directory of organizations and suggestions for further reading **Chemistry** Keith B. Hutton, 2001 Trends in Science is a series of essential readings for anyone who wants to know more about how his or her future will be affected as well the series provides accessible and stimulating material for high school and college students for researchers and librarians *Biomedical Ethics and Decision-Making* Matthew A. Butkus, 2022-02-11 Drawing from clinical experience philosophy psychology and current health law and policy Biomedical Ethics and Decision Making is a detailed survey of persistent issues in health care ethics emphasizing the complexities and nuances of practical decision making and yielding a multifaceted and systematic approach to solving problems As a useful resource for both students and clinicians it includes references for further exploration of ethical issues as well as provocative questions for discussion in classroom and clinical settings As a textbook it stands

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