

Chapter

6

CHEMICAL BONDING

MULTIPLE CHOICE QUESTIONS

- Which of the following molecules has a coordinate covalent bond?
(a) HCl (b) NaCl (c) NH_4Cl (d) AlCl_3 [LHR 05]
- The angle formed in sp hybridization is:
(a) 120° (b) 180° (c) 109.5° (d) 107.5° [GRW 06]
- Ionic compounds are mostly obtained by the combination of group:
(a) 3 and 5 (b) 2 and 5 (c) 4 and 8 (d) 1 and 7 [GRW 07]
- Dipole moment of CO_2 is:
(a) 1.84 D (b) Zero D (c) 0.95 D (d) 2.2 D [GRW 07]
- In sp^2 hybridization, the orbitals are oriented at an angle of:
(a) 109.5° (b) 120° (c) 180° (d) 0° [LHR 07]
- Which of the following species has unpaired electron in anti bonding molecular orbitals?
(a) H_2 (b) He_2 (c) O_2^{+2} (d) N_2^{+2} [MTN 08, 10, SWP, LHR 08]
- A molecular orbital can contain maximum electrons equal to:
(a) One (b) Two (c) Three (d) Four [LHR 08]
- Carbon dioxide and methane have dipole moment value:
(a) Zero and 1.85 D (b) 1.70 D and 1.80 D (c) Both have zero (d) None of these [GRW 08]
- Nobel gases have maximum stability and least reactive because:
(a) They are very safe (b) Their valence shells are complete
(c) They are gases (d) They are present in zero group [GRW 07]
- Which species has unpaired electron in anti-bonding molecular orbits?
(a) O_2^{+2} (b) N_2^{+2} (c) B_2 (d) F_2 [FSD 11, GRW 09, 12]
- Octet rule is not followed in the formation of:
(a) NF_3 (b) CF_4 (c) CCl_4 (d) PCl_5 [LHR 09]
- The most stable elements are:
(a) Halogens (b) Lithium family (c) Noble gases (d) None of these [GRW 09, LHR 14]
- The hybridization of carbon in C_2H_4 is:
(a) sp (b) sp^2 (c) sp^3 (d) not hybridized [LHR 09]
- Positive ions are called:
(a) Cations (b) Anions (c) Molecules (d) Hydrated ions [LHR 11]
- Total number of bonds in C_2H_4 molecule is:
(a) Six (b) Four (c) Five (d) Eight [LHR 10, 11]
- The S.I unit of dipole moment is:
(a) Joule (b) Debye (c) Coulomb meter (d) Nm^{-2} [GRW 11]
- Ionic, covalent and co-ordinate covalent bond are present in:
(a) SO_2 (b) NH_4Cl (c) C_2H_2 (d) H_2O [FSD, GRW 11]
- The highest electronegative element in the periodic table is:
(a) Oxygen (b) Nitrogen (c) Chlorine (d) Fluorine [LHR 10]
- Bond angle between H-S-H bonds is:
(a) 105.5° (b) 107.5° (c) 92° (d) 95° [LHR 10]
- S.I Unit of dipole moment is:
(a) pm (b) Debye (c) mC (d) All [LHR 10]
- The amount of energy released by absorbing electron in the valence shell is:
(a) Ionization energy (b) Electron affinity (c) Electro-negativity (d) Atomization energy [MTN 08, 09, GRW 10]
- The number of electrons shared in SF_6 :
(a) 4 (b) 12 (c) 6 (d) 8 [GRW 10]

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Intelligent Technologies for Interactive Entertainment Martin Clayton, Mauro Passacantando, Marcello Sanguineti, 2024-03-22 This book constitutes the refereed proceedings of the 14th International Conference on Intelligent Technologies for Interactive Entertainment INTETAIN 2023 which was held in Lucca Italy during November 27 2023 The 15 full papers presented in this book were selected from 56 submissions They present novel and innovative work in areas of methods machine learning movement computer based systems architectures software algorithms and devices digital cameras smartphones The papers are grouped in sessions of thematic issues on Games and Game Based learning Motion Capture Sports and Competitions and Interfaces and Applications

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Biology Sylvia S. Mader,2000-07

New York Times Crossword Puzzle Dictionary Outlet,Outlet Book Company Staff,Random House Value Publishing Staff,Rh Value Publishing,1988-12

The Economist ,1999

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Study Guide [for] Charles Seiger,2000

Materials World ,1996

Comfort ,1891

The Bulletin ,1997

Forest & Outdoors and Rod & Gun ,1970

Practical English, a Scholastic Magazine ,1958

High School Chemistry Teachers Magazine ,1967

Essentials of Anatomy and Physiology Charles M. Seiger,Edwin F. Bartholomew,Frederic H.

Martini, 2006 Designed to help students master the topics and concepts covered in the textbook the Study Guide includes a variety of review questions including labeling concept mapping and crossword puzzles that promote an understanding of body systems It is keyed to each chapter's learning objectives and parallels the three level learning system in the textbook

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