

May 2010
SEC Chemistry Paper 1
Marking Scheme

Section A

- 1 (a) Electron -1 ½
- (b) Neutron 0 ½

	Electrons	Neutrons	Protons	Mass number
Atom R	17	18		
Ion Q ²⁻			8	16
Ion T ⁺	2			7
Atom W	17			37

½ x 8

- (ii) R and W Award 1 if **both** correct
Award 0 if one correct
- Total 6 marks**

- 2 (a) (i) Moles carbon = $85.71 / 12 = 7.142$ 1
Moles hydrogen = $14.29 / 1 = 14.29$ 1
Ratio 1C : 2H E.F. = CH₂ 1
If E.F is given as (CH₂)_n deduct ½ mark
- (ii) M.F. = 3 x EF ½
E.F. = C₃H₆ ½
- (b) (i) C₃H₆ + Br₂ → CH₃CHBrCH₂Br accept C₃H₆Br₂
All correct formulae award 2 2
If not all correct award 1
- (ii) Red-orange bromine is decolorized / to colorless ½ x 2
Accept orange-brown
Do not accept brown
Award ½ if the colour is indicated and the answer is "decolorised".
- (iii) Any valid structure for an alkyne, showing triple bond 1
Award 0 if triple bond not shown
Award ½ if triple bond carbon is shown with five bonds
Accept bromoalkene.

Total 8 marks

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Iseb Iseb



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