

Question Number	Acceptable Answers	Reject	Mark
22 (e)	Marking Point 1 London forces/ Dispersion forces Temporary/instantaneous dipole induced dipole forces (1) ALLOW van der Waals forces Marking Point 2 Silane/silicon has more electrons (1) Marking Point 3 Silane has stronger/more London forces (1) Marking Point 4 London forces are weak OR Little energy is required to break London forces (hence both are gases) (1) Ignore references to surface area	Dipole-dipole Induced dipole-induced dipole Hydrogen bonds Larger molar mass/ Great density of electrons/ larger electron cloud Break Si-H or C-H bonds	4

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Steph Warren

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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