

(v) When **Y** is heated with an **alcoholic** solution of potassium hydroxide, the alkene  $C_4H_8$  is formed. What type of reaction occurs to produce the alkene?

(1)

(ib) Separate ethanolic solutions of **X** and **Y** were warmed with water containing dissolved silver nitrate. Describe what would be seen in each case, ignoring any differences in the rates of reaction.

(2)

Observation with **X**

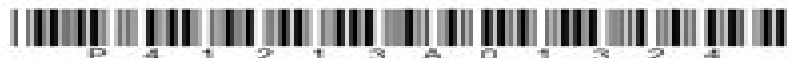
Observation with **Y**

(c) The rates of hydrolysis of primary halogenoalkanes are affected by the strength of the bond between the carbon and the halogen atom.

Is the C—Br bond weaker or stronger than the C—I bond? Explain why the bond strength differs.

(2)

(Total for Question 22 = 12 marks)



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