

Question Number	Acceptable Answers	Reject	Mark
<b>19(a)</b>	200 / $2 \times 10^2$ (ppm)		<b>1</b>

Question Number	Acceptable Answers	Reject	Mark
<b>19(b)(i)</b>	$\text{CH}_3\text{OH(l)} + 3/2\text{O}_2\text{(g)} \rightarrow \text{CO}_2\text{(g)} + 2\text{H}_2\text{O(l)}$ Formulae (1) Balancing and state symbols (1) Allow multiples 2 <sup>nd</sup> mark dependent on 1st	$\text{CH}_3\text{OH(aq)} / \text{(g)}$ $/ 2\text{H}_2\text{O(g)}$	<b>2</b>

Question Number	Acceptable Answers	Reject	Mark
<b>19(b)(ii)</b>	Carbon / C / soot AND carbon monoxide / CO <b>Both needed</b>	Graphite	<b>1</b>

Question Number	Acceptable Answers	Reject	Mark
<b>19(c)(i)</b>	$(150 \times 4.18 \times 15.8) = 9906.6 / 9907 / 9910$ (J) / 9.9066 kJ Ignore sf except 1 sf / Ignore signs here	$\text{kJ mol}^{-1}$	<b>1</b>

Question Number	Acceptable Answers	Reject	Mark
<b>19(c)(ii)</b>	$(0.64/32) = 0.02(00)$ (mol)		<b>1</b>

Question Number	Acceptable Answers	Reject	Mark
<b>19(c)(iii)</b>	$(9.9066/0.0200) = 495.33$ $\Delta H = -495$ (kJ mol <sup>-1</sup> ) Value (1) Sign and 3sf (1) Allow TE from (c)(i) and / or (c)(ii) (answer to (c)(i) in kJ/ answer to (c)(ii)) No 2 <sup>nd</sup> mark if units given are incorrect e.g. kJ mol or kJ/mol <sup>-1</sup>		<b>2</b>

# Edexcel A Level Chemistry January 2014 Wchms

**Katrin Zwirgmaier**



**Edexcel A Level Chemistry January 2014 Wchms:**

## Unveiling the Magic of Words: A Report on "**Edexcel A Level Chemistry January 2014 Wchms**"

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