

## Quiz & Worksheet - Double Displacement Reaction

<http://study.com/academy/practice/quiz-worksheet-double-displacement-reaction.html>

1. What type of reaction is the following?



☐ combustion

☐ gas formation

☐ precipitation

☐ neutralization

2. What are the ions and the correct charges for the following compound:



☐  $\text{Pb}^{2+}$   $\text{Cl}^{2-}$

☐  $\text{Pb}^{2-}$   $\text{Cl}^{2-}$

☐  $\text{Pb}^{+}$   $\text{Cl}^{2-}$

☐  $\text{Pb}^{2+}$   $\text{Cl}^{-}$

3. What are the correct products for the following reaction? (also include the correct coefficient for a balanced reaction)



☐  $\text{Ag}_2\text{SO}_4 + \text{Na}_2\text{NO}_3$

☐  $\text{Ag}_2\text{SO}_4 + \text{NaNO}_3$

☐  $\text{Ag}_2\text{SO}_4 + 2\text{NaNO}_3$

☐  $\text{AgSO}_4 + \text{NaNO}_3$

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# Double Replacement Reactions Lab Answer Key

**Steve Russo, Mike Silver**



## Double Replacement Reactions Lab Answer Key:

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school chemistry or physical science student      **Novice Teacher Action** Anne Liu Kern,2007      **Authentic Practice Chemistry** David G. White,2000      **Introductory Chemistry** Steve Russo, Mike Silver,2002 Steve Russo and Mike Silver turn chemistry into a memorable story that engages readers and provides the context they need to understand and remember core concepts The book builds interesting applications and well designed illustrations into the narrative to get and hold attention then builds confidence with integrated active learning activities Readers make the connections between concepts and the problem solving techniques they need to master as they read The new edition strengthens this conceptual approach and presents additional quantitative techniques in key areas Readers will find enhanced support for quantitative problem solving and more challenging questions at the end of each chapter in addition to the wealth of technology based support on The Chemistry Place tm Special Edition and on The Chemistry of Life CD ROM For college instructors and students

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*Laboratory Experiments for General, Organic & Biochemistry* Frederick A. Bettelheim, Joseph Marvin Landesberg, 1997

**ERDA Energy Research Abstracts** ,1983      **Laboratory Exercises for Preparatory Chemistry** Kathy Dodds Tyner, 1994-06 Laboratory Exercises for Preparatory Chemistry is the perfect complement to a one semester preparatory chemistry laboratory course Tyner s manual emphasizes the application of chemistry and the principles of science to everyday life The labs are directly applicable to the real world and often contain supplemental assignments that illustrate an

application      **Energy Research Abstracts** ,1983      **Exploring Chemistry in Today's World** Kathy L. Tyner,1993 The labs were specifically chosen with several goals in mind a To parallel lecture topics b To demonstrate important chemical principles c To employ the use of techniques of self discovery and the scientific method d To illustrate topics that are of public interest or concern e To encourage the application of chemistry outside the laboratory In keeping with these goals the author has included laboratory assignments that are applicable to the real world or contain supplemental exercises that illustrate an application Where possible commercial products are used such as aspirin antacids etc Each lab begins with written objectives Then in an effort to increase involvement before the lab work begins questions are posed that ask the student a To make predictions about the outcome of the experiment b To formulate a hypothesis c To think about a phenomenon in a specific way d To apply personal experience in answering a questions Pref

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