4.2 Covalent bonding

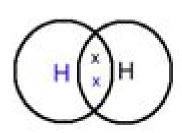
Basic theory

Covalent bonds are formed when electrons from different atoms are shared so that each atom attain a noble gas electron configuration – known as **the octet rule**. The bond is due to the electrostatic attraction between shared pairs of electrons and the positively charged nuclei on either side of the electrons.

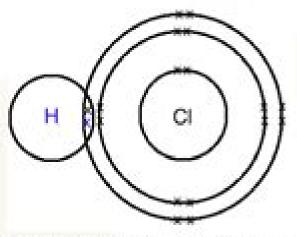
Single covalent bond

Formed when one pair of electrons is shared between two atoms.

Examples include hydrogen and hydrogen chloride.



hydrogen molecule, H2



hydrogen chloride molecule, HCl

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Paul M. S. Monk

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