NAMING COVALENT COMPOUNDS PRACTICE WORKSHEET

A. Write the formulas of the following covalent compounds. 1) Antimony Tribromide 2) Hexaboron Silicide 3) Chlorine Dioxide 4) Hydrogen lodide 5) Iodine Pentafluoride 6) Dinitrogen Trioxide 7) Ammonia 8) Phosphorus Triiodide B. Write the names of the following covalent compounds: 1) P_{*}S₅ 210-3) SeFa 4) Si₂Br₄ 5) SCI. 6) CH₄ 7) B₂Si 8) NF₃

C. Complete the table with the appropriate information.

Name of the Compound	Formula
Dinitrogen Trisulfide	
	CIO ₂
Carbon Diselenide	
Xenon Hexafluoride	
	P4O10

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Peter J. Mikulecky, Chris Hren

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