

Cp physics - Spring Final Review (second semester topics)**Multiple Choice**

Identify the choice that best completes the statement or answers the question.

- _____ 1. Which of the following is a direct cause of a substance's temperature increase?
- Energy is removed from the particles of the substance.
 - Kinetic energy is added to the particles of the substance.
 - The number of atoms and molecules in a substance changes.
 - The volume of the substance decreases.
- _____ 2. What happens to the internal energy of an ideal gas when it is heated from 0°C to 4°C ?
- It increases.
 - It decreases.
 - It remains constant.
 - It is impossible to determine.
- _____ 3. Which of the following is proportional to the kinetic energy of atoms and molecules?
- elastic energy
 - temperature
 - potential energy
 - thermal equilibrium
- _____ 4. As the temperature of a substance increases, its volume tends to increase due to
- thermal equilibrium.
 - thermal expansion.
 - thermal contraction.
 - thermal contraction.
- _____ 5. What is the temperature of a system in thermal equilibrium with another system made up of water and steam at 1 atm of pressure?
- 0°F
 - 273 K
 - 0 K
 - 100°C
- _____ 6. Energy transferred as heat occurs between two bodies in thermal contact when they differ in which of the following properties?
- mass
 - specific heat
 - density
 - temperature
- _____ 7. Which of the following terms describes a transfer of energy?
- heat
 - internal energy
 - temperature
 - kinetic energy
- _____ 8. The use of fiberglass insulation in the outer walls of a building is intended to minimize heat transfer through what process?
- conduction
 - radiation
 - convection
 - vaporization
- _____ 9. How is energy transferred as heat always directed?
- from an object at low temperature to an object at high temperature
 - from an object at high temperature to an object at low temperature
 - from an object at low kinetic energy to an object at high kinetic energy
 - from an object with higher mass to an object of lower mass
- _____ 10. What three properties of a substance affect the amount of energy transferred as heat to or from the substance?
- volume, temperature change, specific heat capacity
 - density, temperature change, specific heat capacity
 - mass, temperature change, specific heat capacity
 - mass, temperature change, latent heat

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