




## Coulombic Attraction

What variables will affect the force of attraction between charged particles?

**Coulombic attraction** is the attraction between oppositely charged particles. For example, the protons in the nucleus of an atom have attraction for the electrons surrounding the nucleus. This is because the protons are positive and the electrons are negative. The attractive force can be weak or strong. In this activity, you will explore the strength of attraction between protons and electrons in various atomic structures.

### Model 1 - Distance and Attractive Force

		Force of Attraction (Newtons)
A		$2.30 \times 10^{-8}$
B		$0.58 \times 10^{-8}$
C		$0.26 \times 10^{-8}$

1. What subatomic particles do these symbols represent in Model 1?



2. Would you expect to observe attraction or repulsion between the subatomic particles in Model 1?

\_\_\_\_\_

3. Consider the data in Model 1.

Write a complete sentence that describes the observed relationship between the distance between the charges and the force of attraction.

\_\_\_\_\_

\_\_\_\_\_

4. If the distance between a proton and electron is 0.50 nm, would you expect the force of attraction to be greater than or less than  $0.26 \times 10^{-8}$  N?

\_\_\_\_\_

5. If two protons are 0.10 nm away from an electron, would you expect the force of attraction to be greater than or less than  $2.30 \times 10^{-8}$  N? \_\_\_\_\_

# Coulombic Attraction Worksheet Key

**S Nieto**



**Coulombic Attraction Worksheet Key:**  
**Science Scope ,1997**

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## **Table of Contents Coulombic Attraction Worksheet Key**

1. Understanding the eBook Coulombic Attraction Worksheet Key
  - The Rise of Digital Reading Coulombic Attraction Worksheet Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Coulombic Attraction Worksheet Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Coulombic Attraction Worksheet Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Coulombic Attraction Worksheet Key
  - Personalized Recommendations
  - Coulombic Attraction Worksheet Key User Reviews and Ratings
  - Coulombic Attraction Worksheet Key and Bestseller Lists
5. Accessing Coulombic Attraction Worksheet Key Free and Paid eBooks
  - Coulombic Attraction Worksheet Key Public Domain eBooks
  - Coulombic Attraction Worksheet Key eBook Subscription Services
  - Coulombic Attraction Worksheet Key Budget-Friendly Options

6. Navigating Coulombic Attraction Worksheet Key eBook Formats
  - ePub, PDF, MOBI, and More
  - Coulombic Attraction Worksheet Key Compatibility with Devices
  - Coulombic Attraction Worksheet Key Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Coulombic Attraction Worksheet Key
  - Highlighting and Note-Taking Coulombic Attraction Worksheet Key
  - Interactive Elements Coulombic Attraction Worksheet Key
8. Staying Engaged with Coulombic Attraction Worksheet Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Coulombic Attraction Worksheet Key
9. Balancing eBooks and Physical Books Coulombic Attraction Worksheet Key
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Coulombic Attraction Worksheet Key
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Coulombic Attraction Worksheet Key
  - Setting Reading Goals Coulombic Attraction Worksheet Key
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Coulombic Attraction Worksheet Key
  - Fact-Checking eBook Content of Coulombic Attraction Worksheet Key
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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