



COVALENT BONDING

Vocabulary Review

Directions: Read each description and write the correct term in the space provided.

an electron pair

electron movement

electron pair

electron pair

electron pair

electron pair

electron pair

electron pair

electron pair

electron pair

electron pair

electron pair

1. A bond formed by the sharing of electron pairs between two atoms.

covalent bond

2. The energy required to break a covalent bond into two free atoms.

bond dissociation energy

3. The energy required to break a covalent bond into two free atoms.

bonding energy level

4. The energy required to break a covalent bond into two free atoms.

electron bond

5. The energy required to break a covalent bond into two free atoms.

electron bond

6. The energy required to break a covalent bond into two free atoms.

electron bond

7. The energy required to break a covalent bond into two free atoms.

electron bond

8. The energy required to break a covalent bond into two free atoms.

electron bond

9. The energy required to break a covalent bond into two free atoms.

electron bond

10. The energy required to break a covalent bond into two free atoms.

electron bond

Covalent Bonding Vocabulary Review

Schyrlet Cameron, Carolyn Craig

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author names.

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on answering the questions The text questions and cases have been updated to reflect the changing influence of microbiology in the world today from the microbiome to new disease outbreaks Ebola and Zika and antibiotic resistance to new biotechnology tools CRISPR Cas

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