### CHEM 120N | EDAPT COVALENT BONDING QUIZ WITH ANSWERS | 2024

Introduction to General Organic and Biological Chemistry with Lab-Mueller > Unit 2

The compound nitrogen trifluoride would have what chemical formula?

N,F

(NF)

NIF.

NE

Which of the pairs of atoms below would be expected to make covalent bonds? Select all that apply:

Calcium and nitrogen Sodium and bromine

Hydrogen and oxygen Carbon and chlorine

In a covalent bond, electrons are:

shared destroyed taken transferred

Which compounds would you expect to contain one or more covalent bonds

| Covalent Bonds                |                                | No Covalent Bonds              |
|-------------------------------|--------------------------------|--------------------------------|
| H <sub>2</sub> O              | S <sub>2</sub> F <sub>10</sub> | Ca <sub>3</sub> P <sub>2</sub> |
| S <sub>8</sub>                | CBr <sub>4</sub>               | CaF <sub>2</sub>               |
| N <sub>2</sub> O <sub>3</sub> |                                | MgO                            |

# **Covalent Bonding Quiz Answers**

**Patrick Vollmar** 

#### **Covalent Bonding Quiz Answers:**

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