| | VEY | Disale | |
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| Name: _ | 10 | Block: | |

Assignment #2 - Compound Names and Formulas Multi-valent ions only

| | ng compounds, all of which have been named using the ic charge is given after the first element (metallic) in Roman |
|---------------------------------|--|
| 1. copper (II) oxide | 11. manganese (III) oxide My 203 |
| 2. mercury (I) oxide +\0,20 | 12. vanadium (II) bromide VB(z |
| 3. gold (III) chloride _AuCl3 | 13. niobium(V) oxide No205 |
| 4. thallium (III) bromide TISV3 | 14. titanium (III) oxide |
| 5. bismuth (V) oxide | 15. titanium (III) nitride TIN |

- 7. uranium (VI) oxide \(\lambda \) \(\O \rangle \) \(\lambda \) \(\O \rangle \) \(\lambda \) \(\O \rangle \) \(\lambda \)
- 8. protectinium (IV) bromide PaBY4 18. tin (II) oxide Sn0
- 9. cerium (III) oxide <u>Ce2()3</u> 19. thulium (II) bromide <u>Tw/Br/2</u>
- 10. arsenic (V) sulphide A52SS 20. copper (II) bromide CUBYZ
- B. Name the compound using the Roman Numeral Method and balance the ionic charges.
- 1. SnCL Tin (IV) Chloride 9. Pdf. Palladium (IV) Fluoride
 2. BiBrs Bismuth (V) Bromide 10. OszO, Osmium (III) Oxide
- 3. POO2 POLONIUM (IV) OXICE 11. MOBEZ MOLYBOLENIUM (II) Bramide
- 4. Pbl. Lead (I) Iodide 12. Vol. Vanadium (V) Chloride
- 5. HgO <u>Hercury (II) Oxide</u> 13. Mn₂O₃ Manganese (III) Oxide 6. HgCl <u>Hercury (II) Chroride</u> 14. Coo <u>Colony (II) Oxide</u>
- 7. Augos Gold (III) Oxide 15. Np2Os NEptunium (III) Oxide
- 8. FeCI: Iron (II) Chloride 16. V2Os Vanadium (V) Oxide

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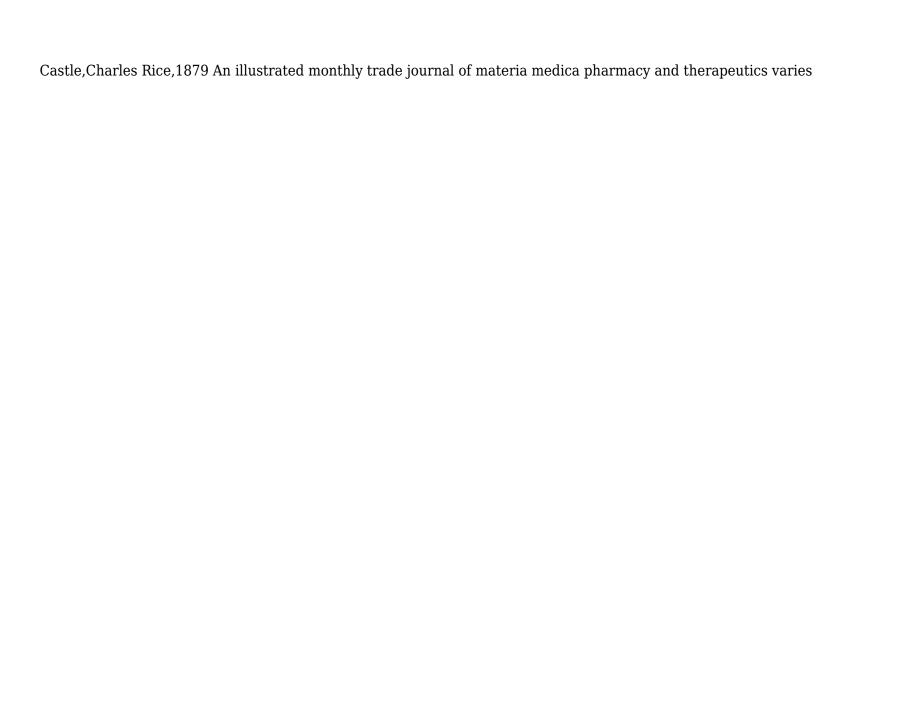
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