# Reference Tables for Physical Setting/EARTH SCIENCE

#### Radioactive Decay Data

RADIOACTIVE ISOTOPE	DISINTEGRATION	HALF-LIFE (years)
Carbon-14	14C →14N	$5.7 \times 10^{3}$
Potassium-40	<sup>40</sup> K → <sup>40</sup> Ar	1.3 × 10 <sup>9</sup>
Uranium-238	236 <sub>U</sub> → 206 <sub>Pb</sub>	4.5 × 10 <sup>9</sup>
Rubidium-87	<sup>87</sup> Rb <sup>87</sup> Sr	4.9 × 10 <sup>10</sup>

## Equations

	stance between foci ength of major axis
	ge in field value distance
Rate of change =	change in value time
Density = mass volume	77 8 28 P. 27 C. S.

#### Specific Heats of Common Materials

MATERIAL	SPECIFIC HEAT (Joules/gram + "C)		
Liquid water	4.18		
Solid water (ice)	2.11		
Water vapor	2.00		
Dry air	1.01		
Basalt	0.84		
Granite	0.79		
Iron	0.45		
Copper	0.38		
Lead	0.13		

## Properties of Water

Heat energy gained during melting 334 J/g
Heat energy released during freezing 334 J/g
Heat energy gained during vaporization 2260 J/g
Heat energy released during condensation 2260 J/g
Density at 3.98°C

#### Average Chemical Composition of Earth's Crust, Hydrosphere, and Troposphere

ELEMENT (symbol)	CRUST		HYDROSPHERE	TROPOSPHERE
	Percent by mass	Percent by volume	Percent by volume	Percent by volume
Oxygen (O)	46.10	94.04	33.0	21.0
Silicon (Si)	28.20	0.88		
Aluminum (Al)	8.23	0.48		
Iron (Fe)	5.63	0.49		
Calcium (Ca)	4.15	1.18		
Sodium (Na)	2.36	1.11		
Magnesium (Mg)	2.33	0.33		
Potassium (K)	2.09	1.42		5-0-0-0-0-0
Nitrogen (N)				78.0
Hydrogen (H)			66.0	
Other	0.91	0.07	1.0	1.0

#### 2011 EDITION

This edition of the Earth Science Reference Tables should be used in the classroom beginning in the 2011–12 school year. The first examination for which these tables will be used is the January 2012 Regents Examination in Physical Setting/Earth Science.

Eurypterus remipos

New York State Fossil

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