



# **Beginner Tutorial For How To Financial Freedom 2025**

**Paul D. Kimmel, Jerry J. Weygandt, Jill E.  
Mitchell**

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web for example take the reaction of lead ii nitrate with hydrochloric acid to form lead ii chloride and nitric acid shown below  
$$\text{Pb(NO}_3)_2 \text{ (aq)} + 2 \text{HCl (aq)} \rightarrow \text{PbCl}_2 \text{ (s)} + 2 \text{HNO}_3$$

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web choose the correct net ionic equation representing the reaction that occurs when solutions of potassium carbonate and copper i chloride are mixed a  $\text{K}_2\text{CO}_3 \text{ aq}$   $2\text{CuCl} \text{ aq}$

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web the net ionic equation is the chemical equation that shows only those elements compounds and ions that are directly involved in the chemical reaction notice that in

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web the net ionic equation for equation 1 is given as equation 3  $\text{s}_2 \text{ aq}$   $\text{Fe}^{2+} \text{ aq}$   $\text{gt}$   $\text{fes s}$  eq 3 8 note that the phase indicator aq is optional on ionic species but is

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