# A QA A-Level Chemistry Paper 1 Unofficial markscheme:

Question	Answer	Debate	Marks
State the meaning of a no transition metal	Central Metal Ion surrounded by co-ordinately bonded ligands	Could you have just said ligands shared lone pair of electrons to central metal ion.  I said transition metal ion coordinately bonded to by ligands. Is that ok? Don't know this question is weird, they should've just asked for the definition of a transition metal. Surely Dative bonding covalent bonds are allowed too?	
PF3 shape and name	Trigonal pyramidal  3 Bonding pairs 1 Lone pair	tetrahedral is allowed no don't think so you might have to put distorted tetrahedral idrk to the same as trig pyramidal Aga can be quite sly so u dont know toh	2
Why does Cr have an oxidation state of 0 in Cr(PF3)6	P has lone pair so it is a ligand Guys for this question, they told us to draw PF3 to understand it has a	I think you hav e to say that PF3 is neutral so Cr has to be zero for overall state to be 0 or something.  Another person agrees	1

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**S Ben Porath** 

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