

**20.1****DESCRIBING ACIDS AND BASES****SECTION REVIEW****Objectives**

- List the properties of acids and bases.
- Name an acid or base when given the formula.

**Key Terms**

- acid
- base

**Part A Completion**

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

Aqueous solutions of \_\_\_\_\_ 1 \_\_\_\_\_ are electrolytes. Acids react with many \_\_\_\_\_ 2 \_\_\_\_\_ to produce \_\_\_\_\_ 3 \_\_\_\_\_ gas. \_\_\_\_\_ 4 \_\_\_\_\_ are compounds that react with acids to form \_\_\_\_\_ 5 \_\_\_\_\_ and a salt.

Acids produce \_\_\_\_\_ 6 \_\_\_\_\_ ions when dissolved in water. There is a standard system for naming acids. For example,  $\text{HBr}$  is called \_\_\_\_\_ 7 \_\_\_\_\_ acid, while  $\text{HNO}_3$  is called \_\_\_\_\_ 8 \_\_\_\_\_ acid.

Bases produce \_\_\_\_\_ 9 \_\_\_\_\_ ions when dissolved in water. Some compounds that are \_\_\_\_\_ 10 \_\_\_\_\_ are named in the same way as other ionic compounds.

1. \_\_\_\_\_

2. \_\_\_\_\_

3. \_\_\_\_\_

4. \_\_\_\_\_

5. \_\_\_\_\_

6. \_\_\_\_\_

7. \_\_\_\_\_

8. \_\_\_\_\_

9. \_\_\_\_\_

10. \_\_\_\_\_

**Part B True-False**

Circlely write *if true* or *false* in the blank line, *AC* for *always true*, *SC* for *always false*, *NT*.

- \_\_\_\_\_ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- \_\_\_\_\_ 12. An acid contains one or more hydroxide ions.
- \_\_\_\_\_ 13. Chemists have a special system for naming bases.

# Describing Acids And Bases Key 20 1

**Xiang Xie**



## **Describing Acids And Bases Key 20 1:**

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science ,2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction

**pH buffering system** Mário Sérgio Piantavini,2025-09-19 Acid Base Balance A Comprehensive Study Guide explores the fundamental principles of pH regulation and buffer systems in human physiology The book presents a historical overview of acid base research from the early 20th century to modern developments followed by an in depth study of key buffering mechanisms the role of lungs and kidneys in homeostasis and the diagnostic use of blood gas analysis It examines metabolic and respiratory disorders their compensatory responses and the complexities of mixed acid base disturbances Special clinical contexts such as cirrhosis hemodialysis and neonatal physiology are addressed highlighting the challenges of maintaining acid base stability in vulnerable populations The book also introduces modern analytical technologies mathematical models and innovative monitoring systems bridging theory and clinical application This guide serves as a valuable resource for students healthcare professionals and researchers seeking a clear yet thorough understanding of acid base physiology and its clinical implications

**Fluid and Electrolyte Therapy, An Issue of Veterinary Clinics of North America: Food Animal Practice** Geof W. Smith,2014-07-28 Fluid and Electrolyte Therapy An Issue of Veterinary Clinics of North America Food Animal Practice

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offers a comprehensive introduction to the structure and function of the human body starting with basic concepts such as the definitions and scopes of anatomy and physiology. The book delves into the levels of structural organization beginning with cells, the building blocks of life, and progresses through tissues, organs, and systems. Each chapter is methodically organized to build upon the previous one, ensuring a logical progression of knowledge. The cellular level of organization explains cell structures, functions, transport mechanisms, division, and intracellular signaling pathways. In the tissue section, the book details the classifications and functional significance of epithelial, muscular, nervous, and connective tissues. The integumentary system chapter highlights the structure and vital protective functions of the skin. The skeletal and muscular systems are examined in detail with emphasis on bone classification, joint articulation, and muscle physiology, including neuromuscular junctions. The book also includes essential insights into the body fluids and blood, outlining components, hematopoiesis, coagulation, and disorders. The lymphatic system section presents the roles of lymph, lymph nodes, and organs in immunity. Further, the peripheral nervous system is thoroughly explored, covering cranial and spinal nerves and the sympathetic and parasympathetic divisions. Special senses are introduced with detailed coverage of the eye, ear, nose, and tongue, along with associated disorders. The cardiovascular system chapter offers a deep dive into heart anatomy, blood flow, vessel structure, and physiological processes like cardiac output and blood pressure regulation. Each system is described in a student-friendly manner, supported by clear terminology and clinical relevance. This book is not just a study guide but a stepping stone toward deeper understanding in the fields of medicine, pharmacy, and allied health sciences. Prentice Hall Chemistry, 2000

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