Name:

mashupmath

# **Balancing Chemical Equations**

Directions: Balance each of the following chemical equations.

- \_\_\_ H<sub>2</sub>PO<sub>4</sub> + \_\_\_ KOH → \_\_\_ K<sub>2</sub>PO<sub>4</sub> + \_\_\_ H<sub>2</sub>O 13
- \_\_ K + \_\_ B<sub>2</sub>O<sub>3</sub> → \_\_ K<sub>2</sub>O + \_\_ B 20
- \_\_\_ HCl + \_\_\_ NaOH → \_\_\_ NaCl + \_\_\_ H<sub>2</sub>O 3)
- \_\_\_ Na + \_\_\_ NaNO<sub>3</sub> -> \_\_\_ Na<sub>2</sub>O + \_\_\_ N<sub>2</sub> 40
- $\_C + \_S_0 \rightarrow \_CS_2$ 50
- \_\_\_ Na + \_\_\_ O<sub>2</sub> -> \_\_\_ Na<sub>2</sub>O 6)
- $N_2 + O_2 \rightarrow N_2O_5$ 7)
- $_{\text{max}}$  H<sub>3</sub>PO<sub>4</sub> +  $_{\text{max}}$  Mg(OH)<sub>2</sub>  $\rightarrow$   $_{\text{max}}$  Mg<sub>3</sub>(PO<sub>4</sub>)<sub>2</sub> +  $_{\text{max}}$  H<sub>2</sub>O 8)
- \_\_\_ NaOH + \_\_\_ H<sub>2</sub>CO<sub>3</sub> → \_\_\_ Na<sub>2</sub>CO<sub>3</sub> + \_\_\_ H<sub>2</sub>O 9)
- \_\_\_ KOH + \_\_\_ HBr → \_\_\_ KBr + \_\_\_ H<sub>2</sub>O 10)
- \_\_ Na + \_\_ O₂ → \_\_ Na₂O 11)
- \_\_\_ Al (OH)<sub>3</sub> + \_\_\_ H<sub>2</sub>CO<sub>3</sub> → \_\_\_ Al<sub>2</sub>(CO<sub>3</sub>)<sub>3</sub> + \_\_\_ H<sub>2</sub>O 12)
- $AI + S_n \rightarrow AI_2S_1$ 13)
- \_\_\_ Cs + \_\_\_ N<sub>2</sub> -> \_\_\_ Cs<sub>2</sub>N 14)
- \_\_ Mg + \_\_ Cl<sub>2</sub> → \_\_ MgCl<sub>2</sub> 15)
- 16) \_\_\_ Rb + \_\_\_ RbNO₁ → \_\_\_ Rb₂O + \_\_\_ N₂
- 17) \_\_C<sub>6</sub>H<sub>6</sub> + \_\_O<sub>2</sub> → \_\_CO<sub>2</sub> + \_\_H<sub>2</sub>O
- 18) \_\_\_ N<sub>2</sub> + \_\_\_ H<sub>2</sub> → \_\_\_ NH<sub>3</sub>

# **Balancing Chemical Equations Worksheet Teacher Web Answers**

**M Tight** 

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