#### **Balancing Chemical Reactions**

#### Balance the following equations:

4. 
$$N_2 + O_2 \rightarrow N_2O_5$$

5. 
$$_H_3PO_4 + _Mg(OH)_2 \Rightarrow _Mg_3(PO_4)_2 + _H_2O$$

7. \_\_\_H<sub>2</sub> + \_\_\_O<sub>2</sub> 
$$\rightarrow$$
 \_\_\_H<sub>2</sub>O<sub>2</sub>

10. 
$$Cs + N_2 \rightarrow Cs_2N$$

11. 
$$\underline{\hspace{1cm}}$$
 Mg +  $\underline{\hspace{1cm}}$  Cl<sub>2</sub>  $\Rightarrow$   $\underline{\hspace{1cm}}$  MgCl<sub>3</sub>

12. \_\_\_Rb + \_\_\_RbNO<sub>8</sub> 
$$\Rightarrow$$
 \_\_\_Rb<sub>2</sub>O + \_\_\_N<sub>2</sub>

13. \_\_\_C\_H\_0 + \_\_\_O\_
$$\rightarrow$$
 \_\_\_CO\_ $\rightarrow$  + \_\_\_H\_0O

15. \_\_ 
$$C + __ O_2 \rightarrow __ CO_2$$

16. 
$$C_2H_0 + C_2 - C_2 + H_0O$$

17. 
$$\underline{\hspace{1cm}}$$
 ii +  $\underline{\hspace{1cm}}$  AlCl<sub>3</sub>  $\Rightarrow$   $\underline{\hspace{1cm}}$  LiCl +  $\underline{\hspace{1cm}}$  Al

$$10. \underline{\hspace{1cm}} Rb + \underline{\hspace{1cm}} P \Rightarrow \underline{\hspace{1cm}} Rb_S P$$

19. 
$$Al(OH)_1 + H_1SO_1 \Rightarrow Al_2(SO_1)_1 + H_2O_1$$

20. Rb + 
$$S_6 \rightarrow Rb$$
, S

21. 
$$CaSO_4 + KOH \rightarrow Ca(OH)_2 + K_2SO_4$$

23. 
$$Al_4C_8 + _H_2O \Rightarrow _CH_4 + _Al(OH)_8$$

24. \_\_Ba + \_\_IINO<sub>3</sub> 
$$\Rightarrow$$
 \_\_II<sub>2</sub> + \_\_Ba(NO<sub>3</sub>)<sub>3</sub>

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