

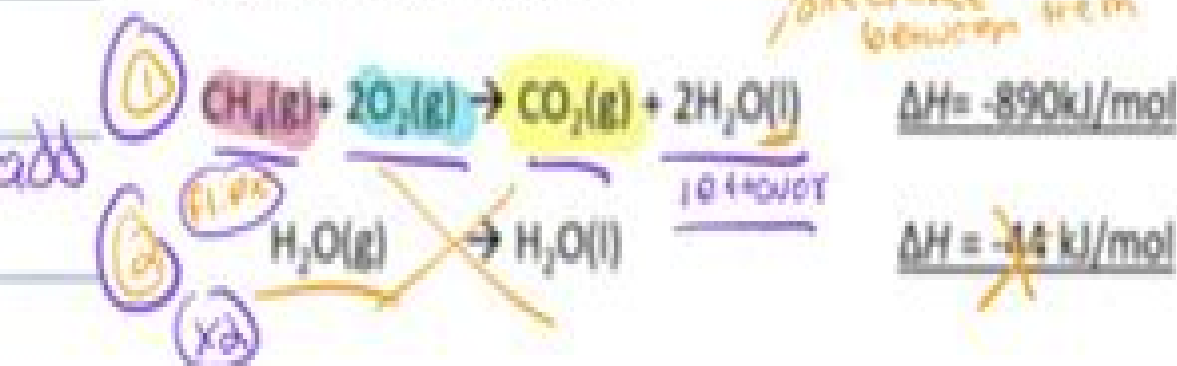
What is the heat of reaction for the following equation:

→ the system

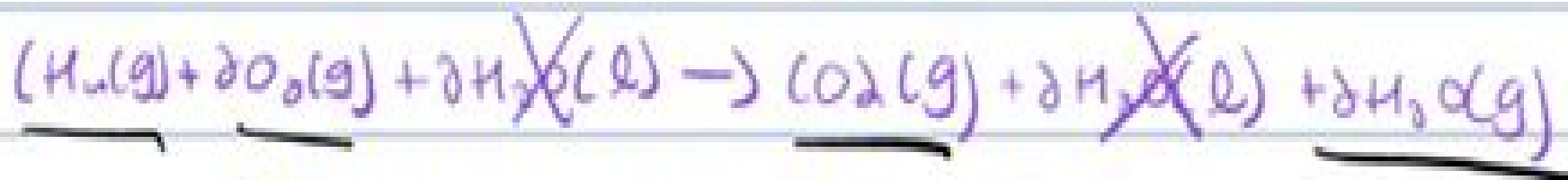


Using the following data:

→ difference between them



$\Delta H = +88 \text{ kJ/mol}$



$-890 \text{ kJ} + 88 = \Delta H = -802 \text{ kJ/mol}$

Calculating Heats Of Reaction Section Review Key

Arnold B. Loebel



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