

Dilution Calculations

What is the molarity (M) of a solution prepared by diluting 0.180L of 0.600 M HNO_3 to 0.540 L?

Prepare a table:

$$M_1 = 0.600 \text{ M} \quad V_1 = 0.180 \text{ L}$$

$$M_2 = ? \quad V_2 = 0.540 \text{ L}$$

Solve dilution equation for unknown and enter values:

$$M_1 V_1 = M_2 V_2$$

$$M_2 = \frac{M_1 V_1}{V_2} = \frac{(0.600 \text{ M})(0.180 \text{ L})}{0.540 \text{ L}} = 0.200 \text{ M}$$

Calculating Molarity By Dilution Answers

John Sylvester Finlayson



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Mathematics for the Clinical Laboratory E-Book Lorraine J. Doucette, 2020-06-18 Selected for Doody's Core Titles 2024 with Essential Purchase designation in Laboratory Technology Master the skills you'll need to perform accurate clinical laboratory calculations Mathematics for the Clinical Laboratory 4th Edition demonstrates the calculations used in the analysis of test specimens It begins by explaining basic mathematical principles and then covers the types of calculations needed in specific areas of the clinical lab including urinalysis hematology and microbiology Finally it focuses on the statistical calculations used in quality assurance and quality control Step by step examples reinforce your understanding and calculation templates and practice problems ensure that you make correct calculations every time Step by step examples explain basic mathematical principles and show you exactly how to perform each type of calculation Sample problems with answers can also be used as templates for solving laboratory calculations Practice problems at the end of each chapter provide a self assessment tool helping you determine what you need to review Summaries of important formulas are included at the end of the text's major sections Coverage of statistical calculations includes standard deviation as well as calculations associated with quality assurance and quality control Quick tips and notes make it easier to understand and remember pertinent information Learning objectives at the beginning of each chapter provide measurable outcomes to achieve by completing the chapter material Full color design includes 100 illustrations Useful appendix of Greek symbols provides a quick reference to turn to when studying Glossary at the back of the textbook includes definitions of important mathematical terms New Updated content and calculations reflect the latest procedures used in today's laboratories

Pharmaceutical Calculations Michalakis Savva, 2019-10-17 Pharmaceutical Calculations A Conceptual Approach is a book that combines conceptual and procedural understanding for students and will guide you to master prerequisite skills to carry out accurate compounding and dosage regimen calculations It is a book that makes the connection between basic sciences and pharmacy It describes the most important concepts in pharmaceutical sciences thoroughly accurately and consistently through various commentaries and activities to make you a scientific thinker and to help you succeed in college and licensure exams Calculation of the error associated with a dose measurement can only be carried out after understanding the concept of accuracy versus precision in a measurement Similarly full appreciation of drug absorption and distribution to tissues can only come about after understanding the process of transmembrane passive diffusion Early understanding of these concepts will allow reinforcement and deeper comprehension of other related concepts taught in other courses More weight is placed on the qualitative understanding of fundamental concepts like tonicity vs osmotic pressure diffusion vs osmosis crystalloids vs colloids osmotic diuretics vs plasma expanders rate of change vs rate constants drug accumulation vs drug fluctuation loading dose vs maintenance dose body surface area BSA vs body weight BW as methods to adjust dosages and much more before considering other quantitative problems In one more significant innovation the origin and physical significance of all

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Chemical Principles, Properties, and Reactions Kenneth L. Henold, Frank Walmsley, 1984

Student Study Guide and Solutions Manual to Accompany General, Organic, and Biochemistry Katherine J. Denniston, Denniston Katherine, 2006

Essential Laboratory Mathematics Catherine W. Johnson, Daniel L. Timmons, Pamela E. Hall, 2009-12-03 This hands on manual with pedagogical features that draw the learner into the content offers clear and complete coverage of the mathematical topics most often used in today's clinical and medical laboratories Furthermore it provides a solid foundation for subsequent courses in the laboratory sciences The first two chapters present a review of basic mathematical concepts The remainder of the book provides students with a realistic means to build on previously learned concepts both mathematical and scientific to refine their mathematical skills and to gauge their mastery of those skills Outstanding features Each chapter opens with an outline objectives and key terms Key terms highlighted within the text are listed and defined in the glossary Margin problems and practice problem sets provide the chance to gain immediate proficiency Laboratory exercises and review problems allow students to apply what they ve

learned and assess their understanding and progress A special calculator icon signals explanations of calculator use for a particular mathematical function Study hints Keys to Success offer practical suggestions and guidance for maximizing achievement The workbook design enables users to solve problems and take notes directly on the pages

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Basic Biochemical Calculations John Sylvester Finlayson,1969

Chemistry James E. Brady,Fred Senese,2004-02-04 Publisher Description

Mathematics for the Clinical Laboratory Lorraine J. Doucette,1997 This comprehensive practical text shows you how to perform the clinical calculations used in each area of the laboratory to ensure accurate results for patients

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Student Study Guide and Solutions Manual to accompany General Organic and Biological Chemistry, 1e Kenneth W. Raymond,2005-10-07 Finally readers have a shorter less intimidating introduction to general organic and biological chemistry Not only is Raymond s text concise it also takes an integrated approach to presenting important topics in a way that makes the material easier to understand In this approach similarities can be exploited and concepts reinforced The result is that readers see the strong connections that exist between these three branches of chemistry

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