Name: _____

Score: ____

Net Ionic Equations

Complete and balance the following equations. Show the ionic and net ionic forms of the equations.

1) K₃PO₄(aq.) + AI(NO₃)₃(aq.) -

lonic:

Net ionic:

2) Bel₂(aq.) + Cu₂SO₄(aq.)

Ionic:

Net ionic:

3) $AgNO_3(aq.) + MgI_2(aq.)$

lonic:

Net ionic:

4) HBr(aq.) + Pb(ClO₄)₂(aq.)

Ionic:

Net ionic:

5) $K_2(C_2O_4)(aq.) + CaCl_2(aq.)$

lonie:

Net ionic:

6) $Na_2CO_3(aq.)$ + $FeCl_2(aq.)$

lonic:

Net ionic:

7) $Ni(NO_3)_2(aq.) + NaOH(aq.)$

lonic:

Net ionic:

8) $Pb(NO_3)_2(aq.) + H_2SO_4(aq.)$

lonic:

Net ionic:

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Crispin Collins

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