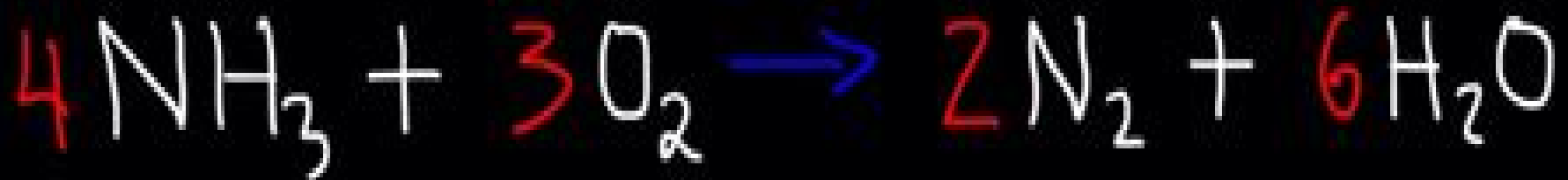


Mole Ratio Problems



$$\frac{11 \text{ mol } \cancel{\text{NH}_3}}{1} \times \frac{3 \text{ mol } \text{O}_2}{4 \cancel{\text{ mol } \text{NH}_3}} = \boxed{8.25 \text{ mol } \text{O}_2}$$

Balancing Equation Mole To Mole Ratios Answer

Xiang Xie



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