

Chemical and Physical Changes and Properties

Purpose

Based on experimental data, determine the occurrence of a chemical change.

Introduction

Text Reference: Chapter 1, *Chemistry, A Molecular Approach*, Nivaldo Tro, 4th ed.

Matter can experience two types of changes:

- A physical change is a change in physical properties (size, mass, state of matter, etc) without a change in chemical composition.
- A chemical change involves the rearrangement of atoms of one or more substances and a change in chemical properties or composition, resulting in the formation of at least one new substance. All chemical changes are accompanied by one or more physical change, along with energy exchanges.

A common example of a physical change is a change in state of the substance. The appearance of a substance changes when you melt, freeze, boil, condense or sublimate it. This change is due to the addition or removal of heat. The composition of the substance remains the same, even if it looks very different. When heat is added to ice, it starts to melt and then the appearance is completely different, from a clear solid to a transparent liquid. If we continue adding heat, the liquid starts to vaporize, and the water now is steam. Even when the ice does not look very similar to steam, the composition is the same: it is water. Cutting, tearing, shattering, and grinding are other examples of physical changes. These may be irreversible, however there is no change in composition.

In a chemical change, a change in the composition occurs. For example, when the Statue of Liberty was erected in 1886, the copper metal was orange in color. However, by 1900, it had turned a turquoise color. The copper underwent a chemical change, going from metallic copper to a mixture of copper oxides, copper carbonates, copper sulfates, and copper chlorides due to exposure to various chemicals in the air and rain.

Certain observations that indicate a chemical change include:

- *Formation of precipitate.* It is the formation of an insoluble solid when two reactants are mixed. It may be a fine suspension or it may form large particles that quickly settle down. It may occur immediately, or it may take some time to settle down in order to be observed.
- *Color change.* It is the observation of a new color in the substance(s) that we are using. When a clean copper wire is heated in a burner flame, the appearance of the copper changes from coppery metallic to glowing red. Finally, its final appearance by cooling is a black solid. It is a new substance; copper oxide. However, when a clean platinum wire

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Table of Contents Chm2045l Lab Manual

1. Understanding the eBook Chm2045l Lab Manual
 - The Rise of Digital Reading Chm2045l Lab Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Chm2045l Lab Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Chm2045l Lab Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Chm2045l Lab Manual
 - Personalized Recommendations

- Chm2045l Lab Manual User Reviews and Ratings
- Chm2045l Lab Manual and Bestseller Lists
- 5. Accessing Chm2045l Lab Manual Free and Paid eBooks
 - Chm2045l Lab Manual Public Domain eBooks
 - Chm2045l Lab Manual eBook Subscription Services
 - Chm2045l Lab Manual Budget-Friendly Options
- 6. Navigating Chm2045l Lab Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Chm2045l Lab Manual Compatibility with Devices
 - Chm2045l Lab Manual Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Chm2045l Lab Manual
 - Highlighting and Note-Taking Chm2045l Lab Manual
 - Interactive Elements Chm2045l Lab Manual
- 8. Staying Engaged with Chm2045l Lab Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Chm2045l Lab Manual
- 9. Balancing eBooks and Physical Books Chm2045l Lab Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Chm2045l Lab Manual
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Chm2045l Lab Manual
 - Setting Reading Goals Chm2045l Lab Manual
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Chm2045l Lab Manual
 - Fact-Checking eBook Content of Chm2045l Lab Manual

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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